

Universidad Nacional Autónoma de México

Química Orgánica III (1506)

Laboratorio

Semestre 2025 - 2

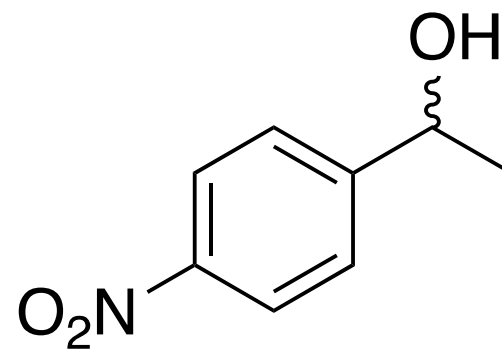


M. en C. Arturo García Zavala

Práctica 1

**Nitrocompuestos I**

**Reducción quimioselectiva de  
4-nitroacetofenona**



19/2/2025

La mayoría de las moléculas orgánicas contienen múltiples grupos funcionales, los cuales pueden reaccionar de diversas maneras.

La **selectividad** se refiere a qué grupo funcional reaccionará, en qué sitio ocurrirá la reacción y de qué manera se llevará a cabo.

Quimioselectividad

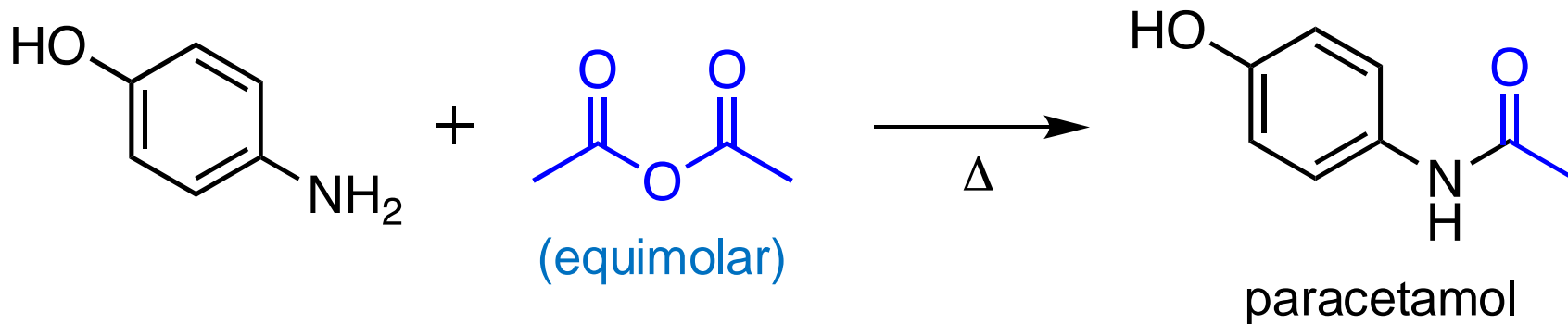
Regioselectividad

Estereoselectividad

# Quimioselectividad

La quimioselectividad es la reacción preferencial de un reactivo químico con uno de dos o más grupos funcionales diferentes.

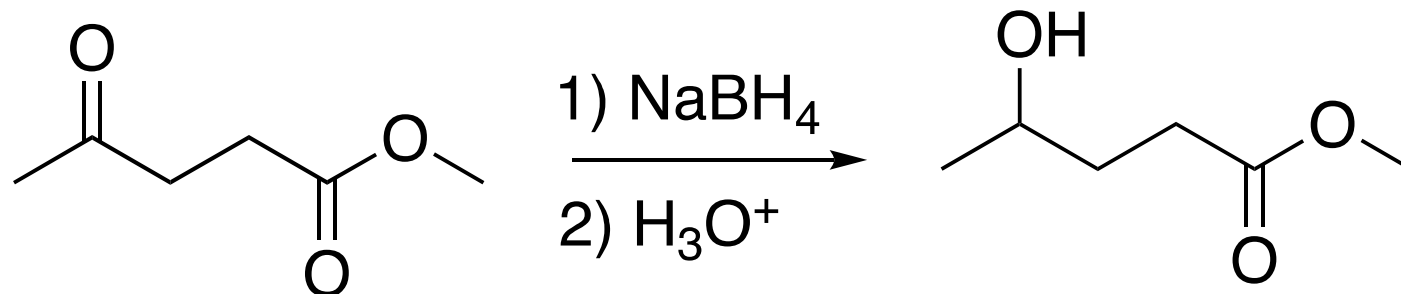
**Acetilación quimioselectiva del 4-aminofenol para la obtención de paracetamol.**



'chemoselectivity (chemoselective)' in *IUPAC Compendium of Chemical Terminology*, 5th ed. International Union of Pure and Applied Chemistry; 2025. Online version 5.0.0, 2025.

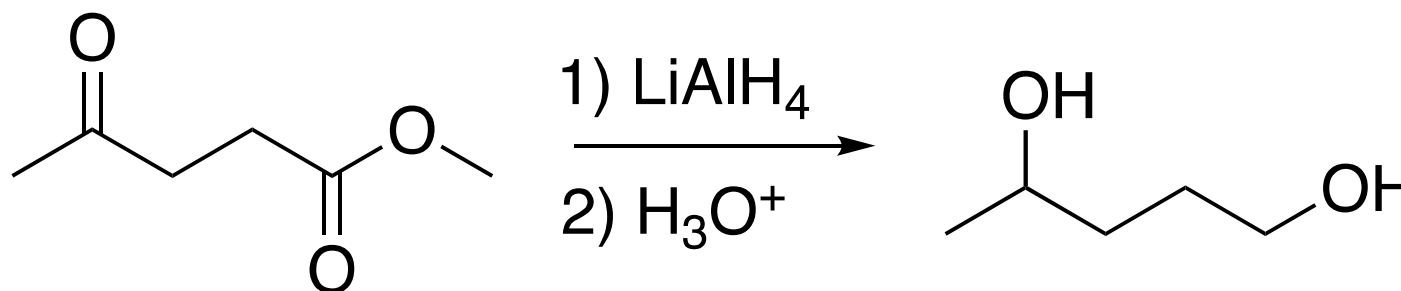
<https://doi.org/10.1351/goldbook.C01051>

## Reducción quimioselectiva



¡El carbono del éster no es muy electrofílico!

## Reducción no quimioselectiva

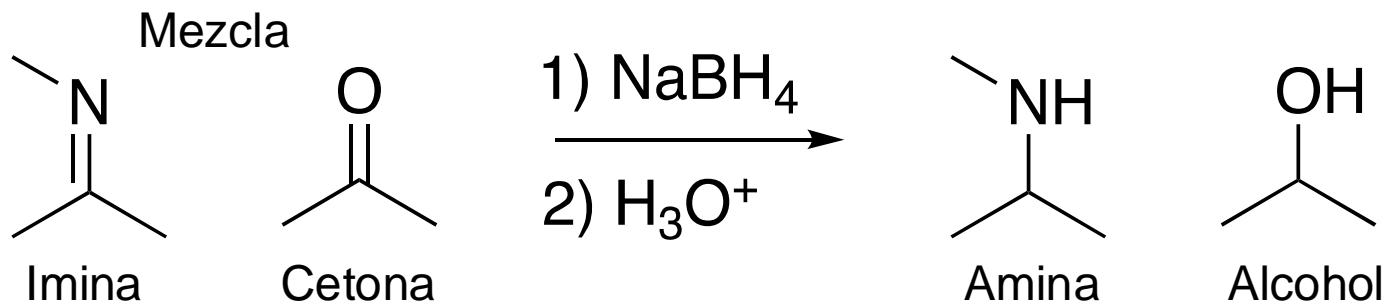


$\text{LiAlH}_4$  es un reductor muy fuerte

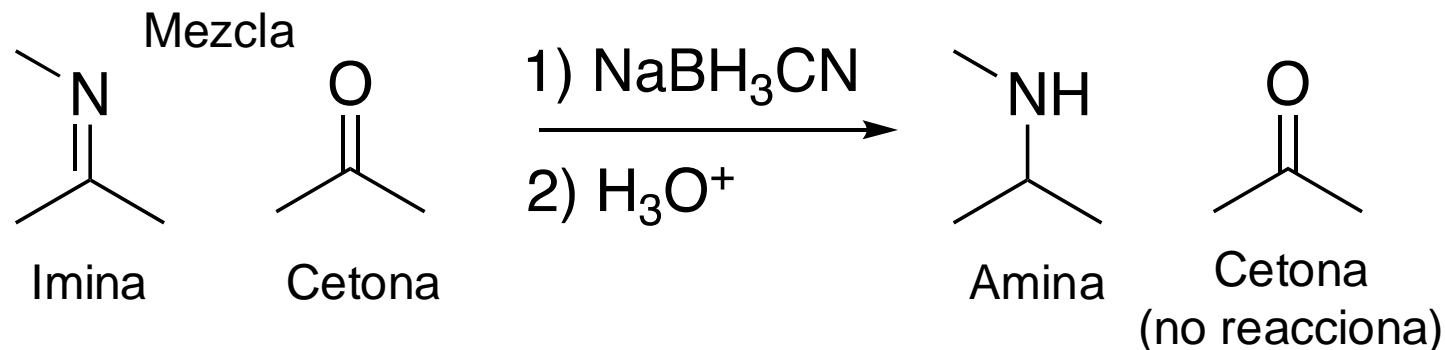


# Cianoborohidruro de sodio como reductor quimioselectivo de iminas

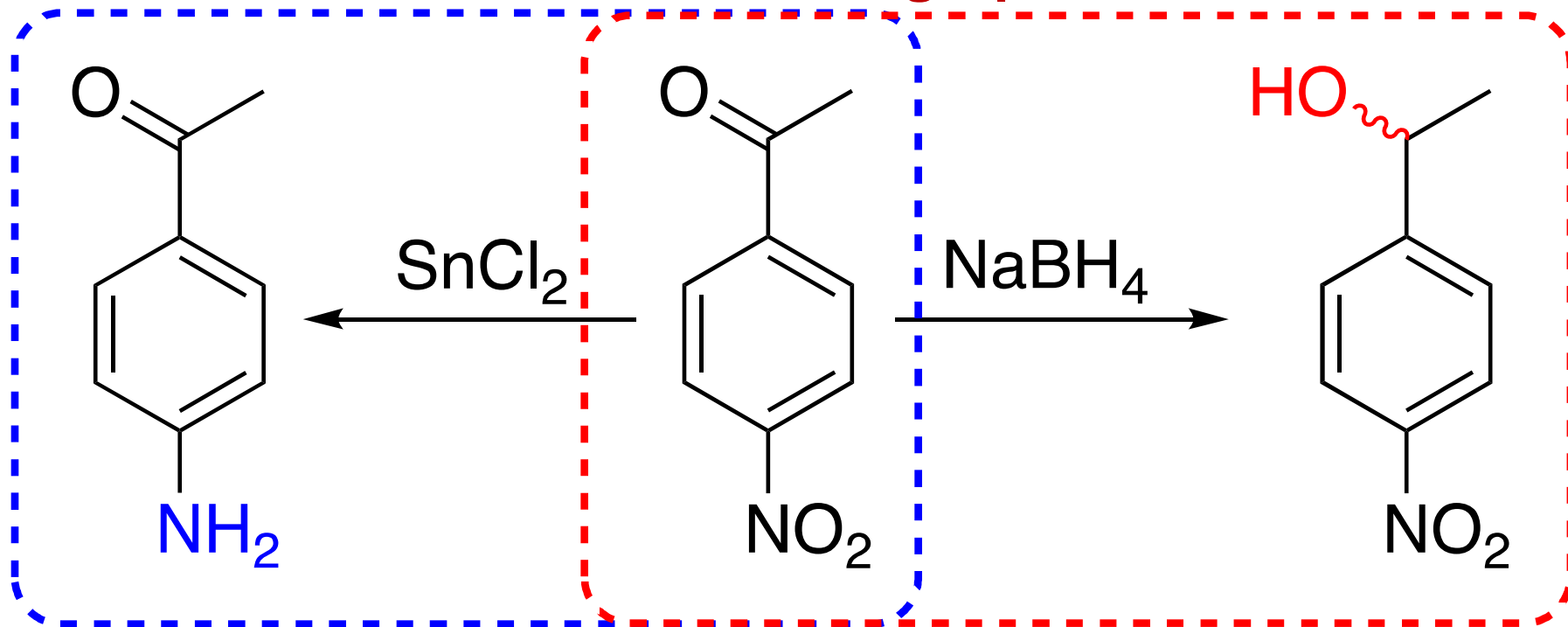
## Borohidruro de sodio reduce iminas y cetonas



## Cianoborohidruro de sodio solo reduce iminas



## Reducción quimioselectiva del grupo carbonilo

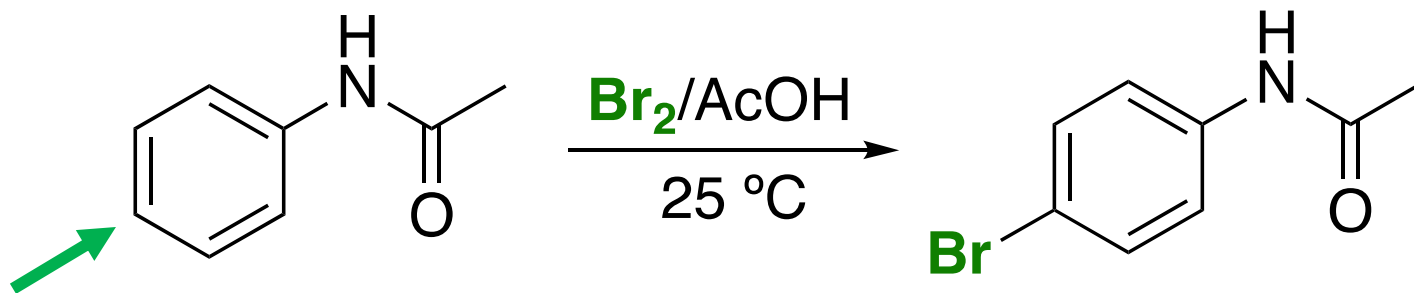


Reducción quimioselectiva del grupo nitro

# Regioselectividad

Una reacción regioselectiva es aquella en la que una dirección de formación o ruptura de enlaces ocurre de manera preferente sobre todas las demás posibles.

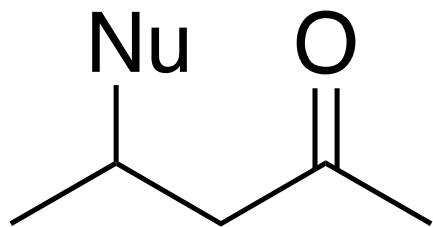
## Bromación regioselectiva de benzamida.



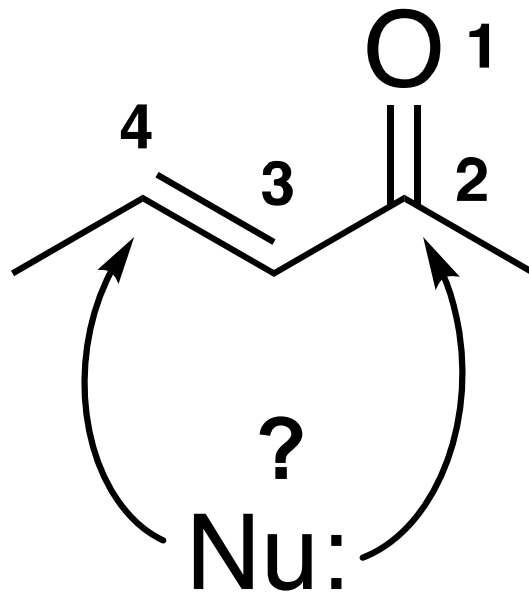
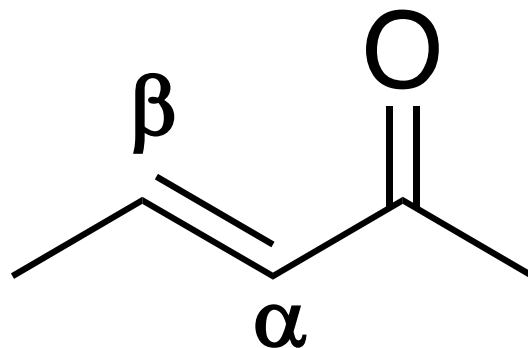
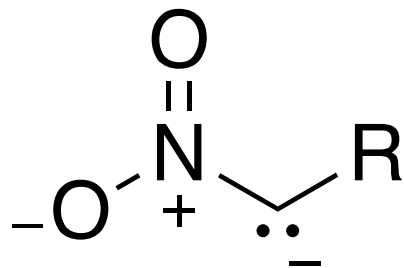
'regioselectivity (regioselective)' in *IUPAC Compendium of Chemical Terminology*, 5th ed. International Union of Pure and Applied Chemistry; 2025. Online version 5.0.0, 2025. <https://doi.org/10.1351/goldbook.R05243>

# Adiciones a compuestos carbonílicos $\alpha,\beta$ -insaturados (aceptores de Michael).

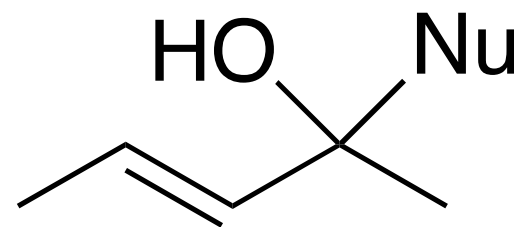
## Adición conjugada (Adición 1,4)



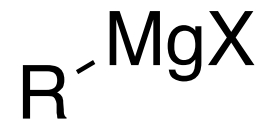
Nucleófilos blandos



## Adición directa (Adición 1,2)



Nucleófilos duros





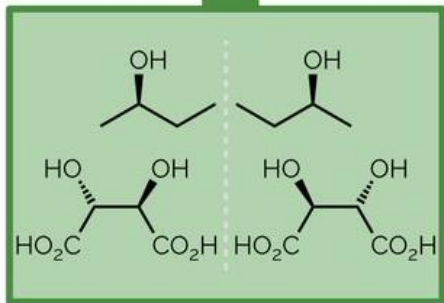
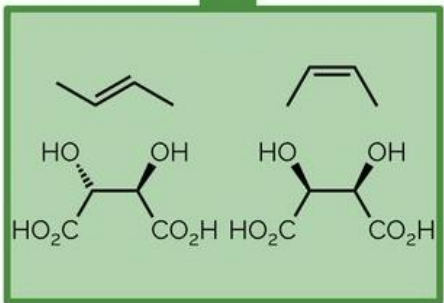
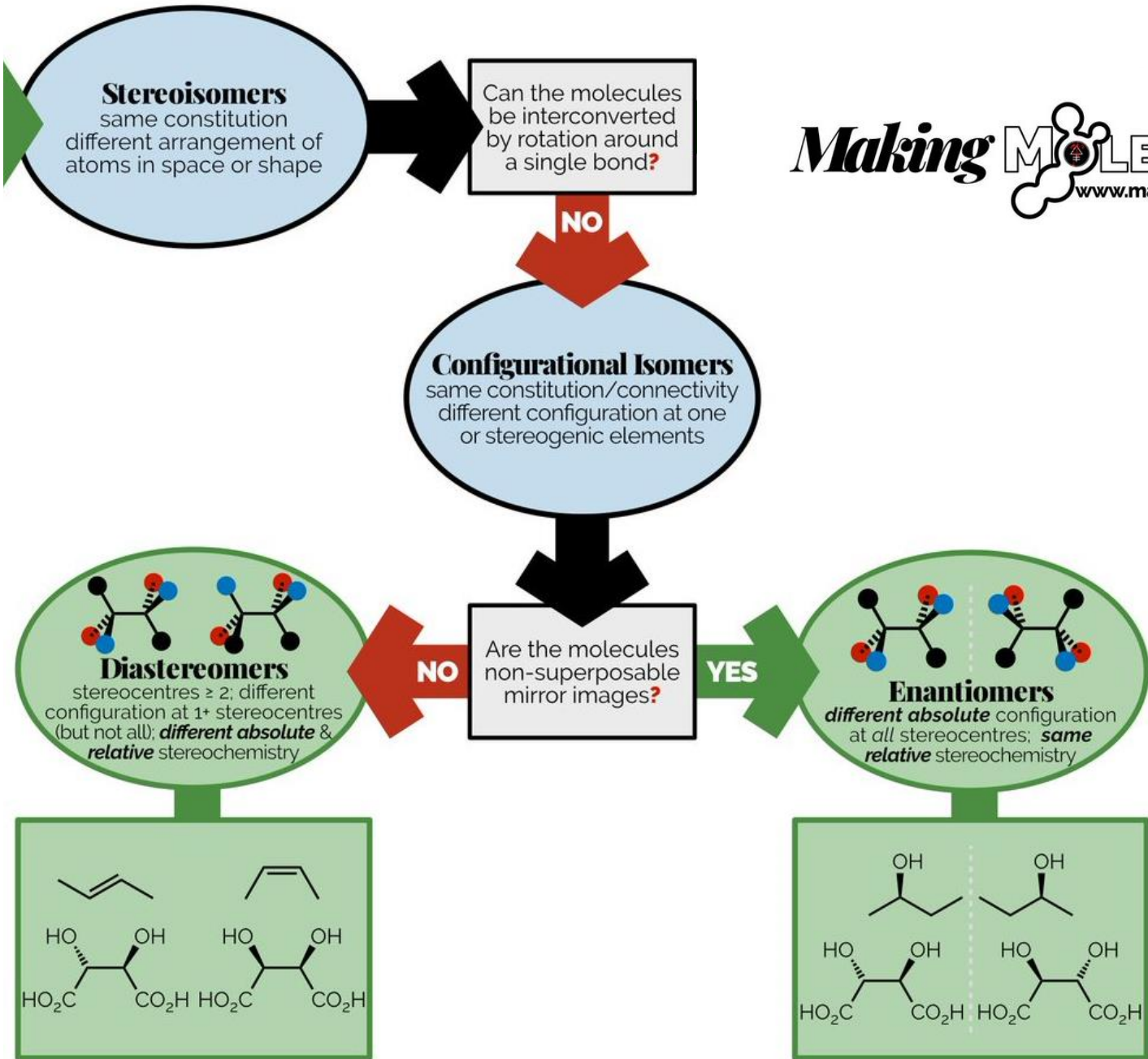
# Estereoselectividad

La formación preferencial de un estereoisómero sobre otro en una reacción química.

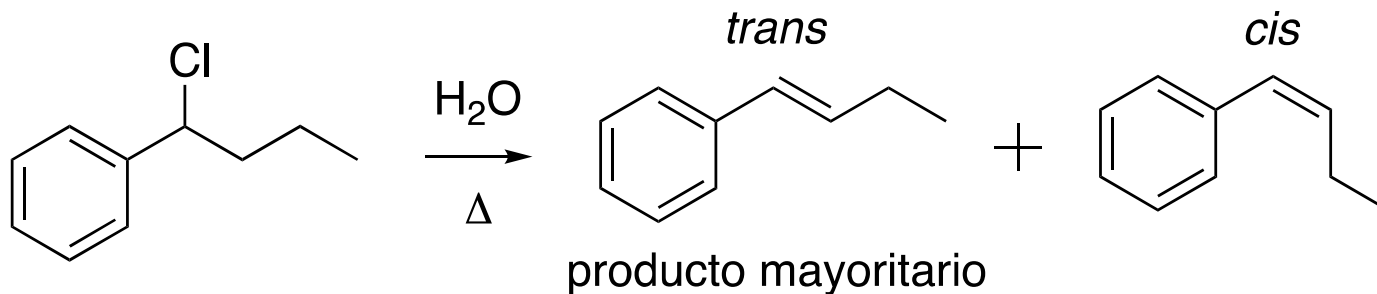
Cuando los estereoisómeros son enantiómeros, este fenómeno se denomina enantioselectividad.

Cuando son diastereoisómeros, se denomina diastereoselectividad.

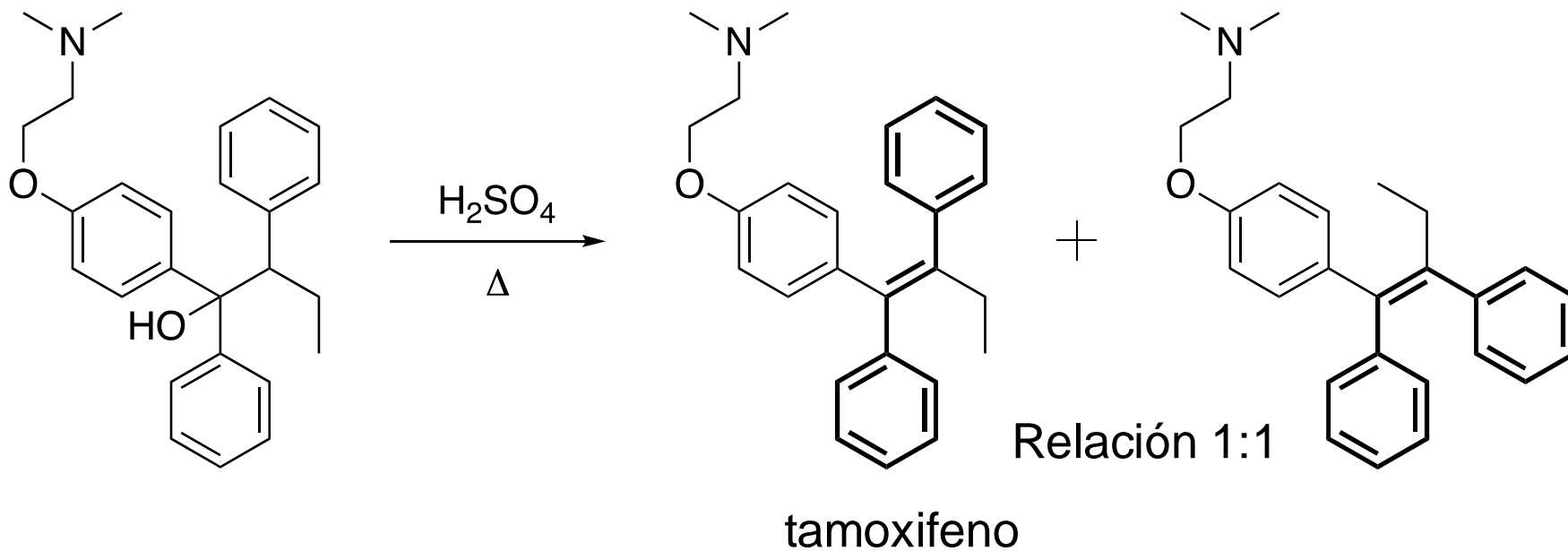
'stereoselectivity' in *IUPAC Compendium of Chemical Terminology*, 5th ed. International Union of Pure and Applied Chemistry; 2025. Online version 5.0.0, 2025. <https://doi.org/10.1351/goldbook.S05991>



# Las eliminaciones de primer orden (E1) son estereoselectivas



# Deshidratación no estereoselectiva



# Otros agentes reductores

## Sustratos

**Cloruro de ácido**

**Amida**

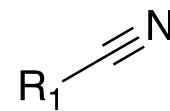
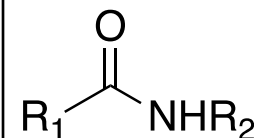
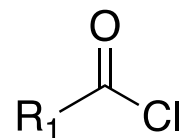
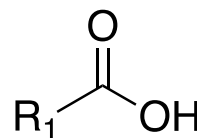
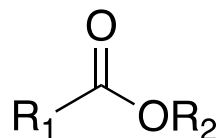
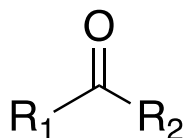
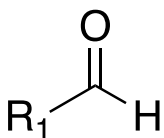
**Nitrilo**

**Aldehído**

**Cetona**

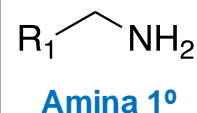
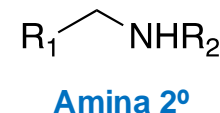
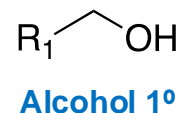
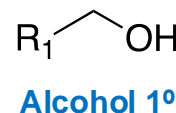
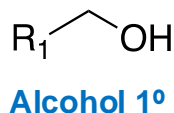
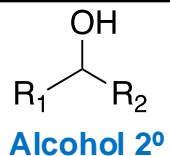
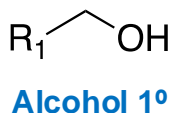
**Éster**

**Ácido**

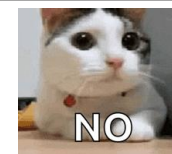
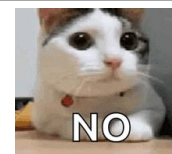
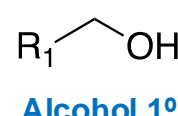
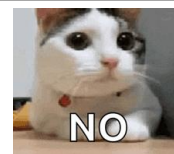
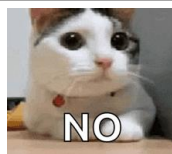
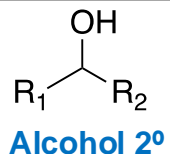
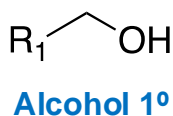


Agentes reductores

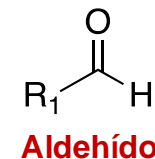
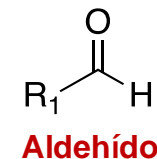
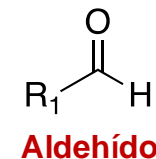
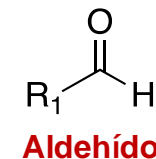
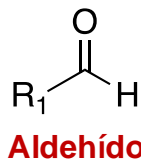
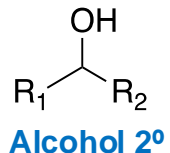
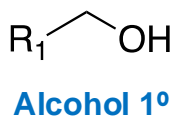
$\text{LiAlH}_4$



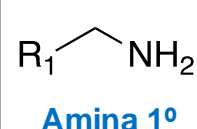
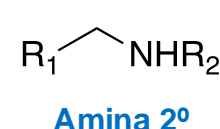
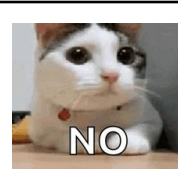
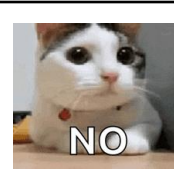
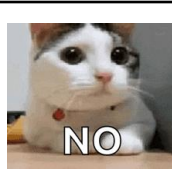
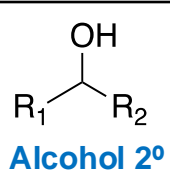
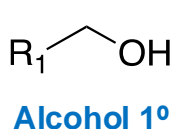
$\text{NaBH}_4$



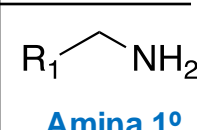
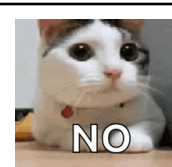
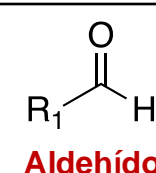
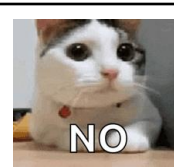
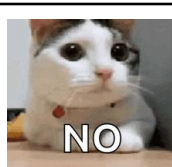
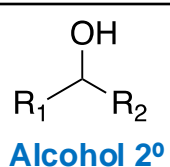
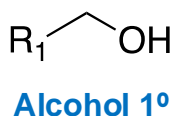
DIBAL-H



$\text{H}_2$ ,  
Ni Raney

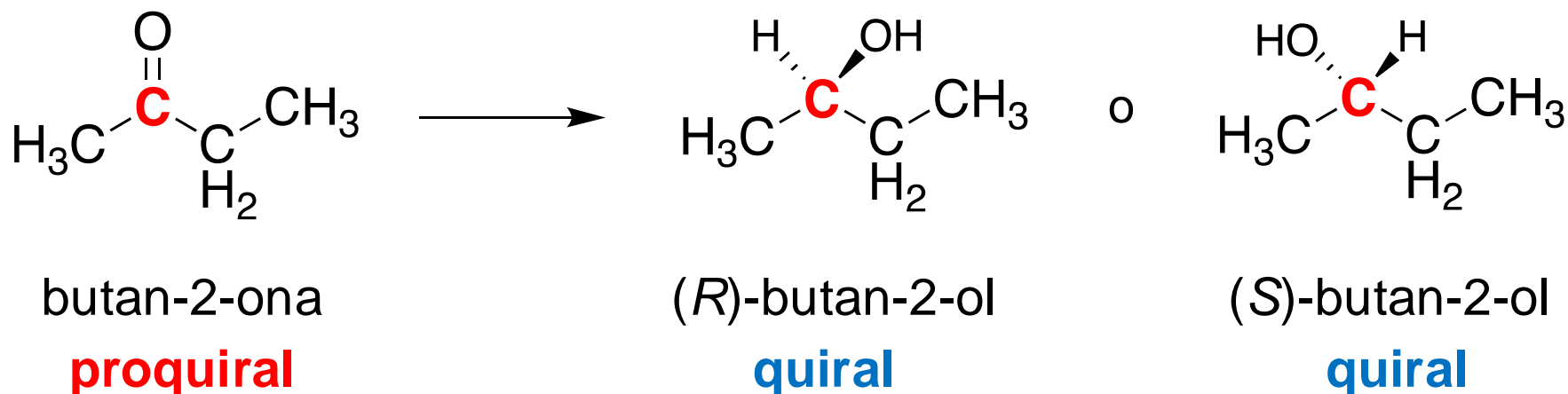


$\text{H}_2$ , Pd/C

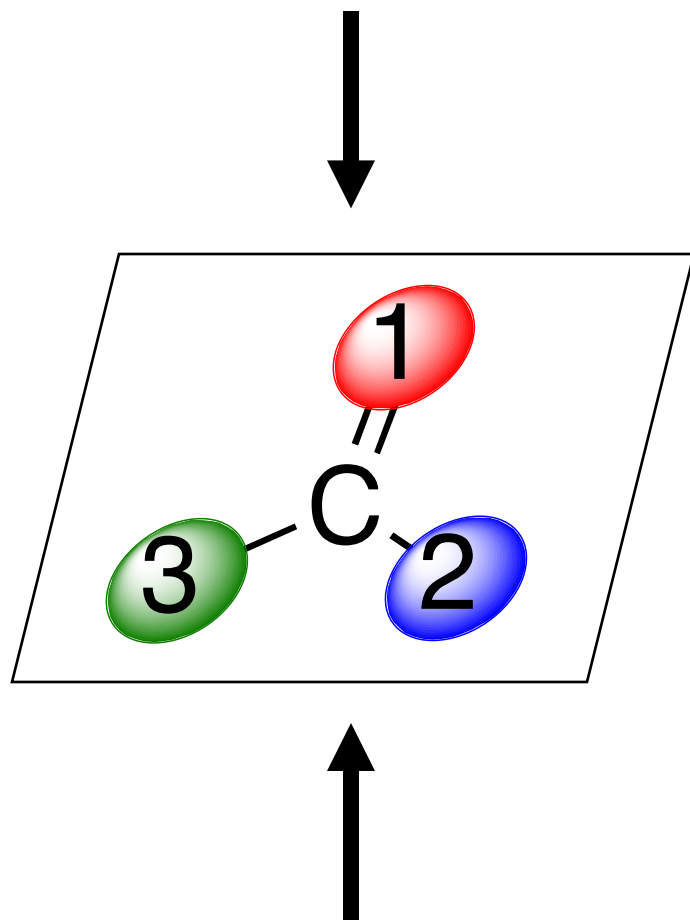


# Proquiralidad

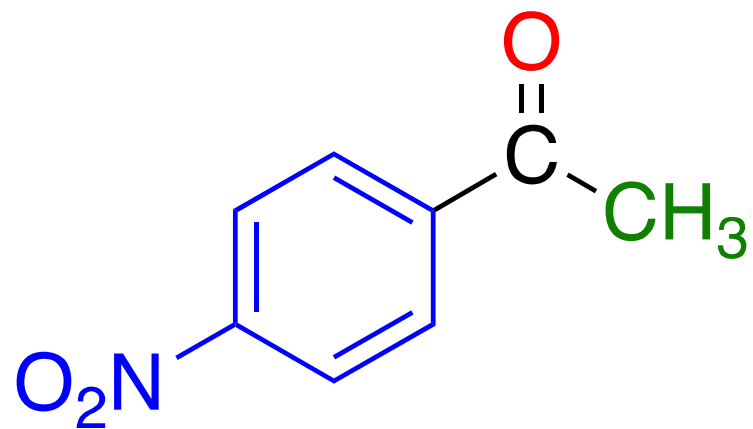
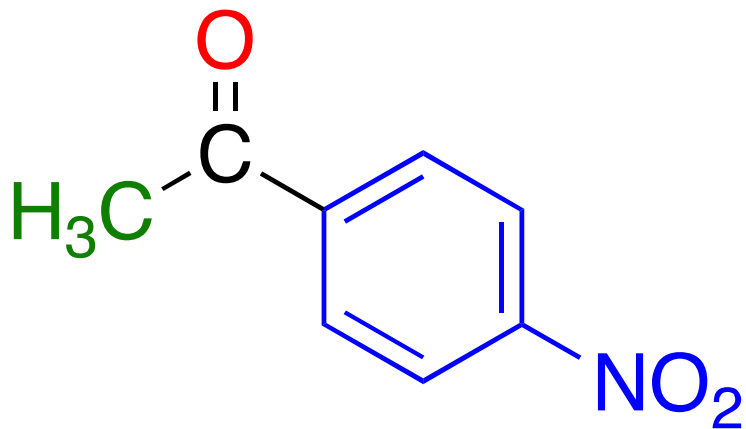
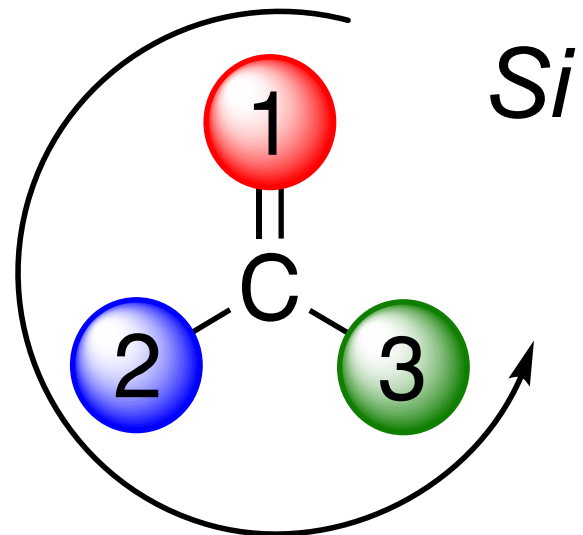
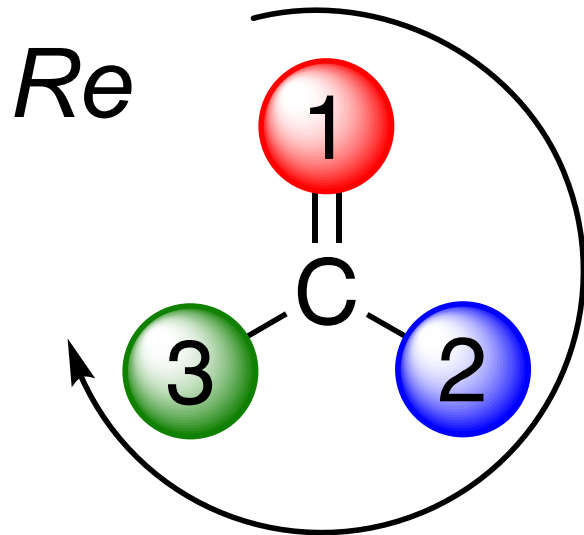
Molécula o entidad aquiral que contiene un sistema trigonal y que puede volverse quiral al adicionarse un nuevo átomo o grupo aquiral.

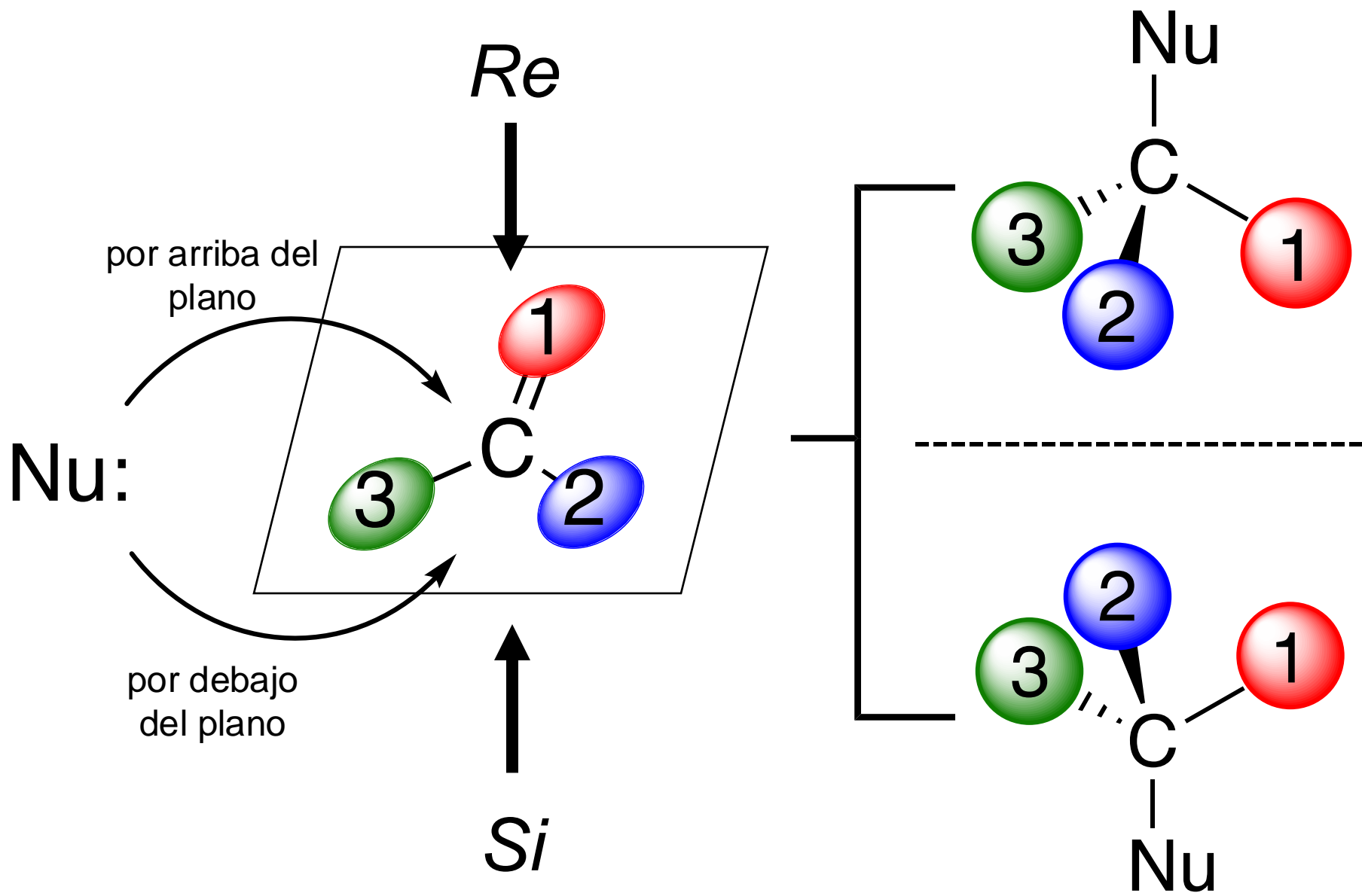


'prochirality' in *IUPAC Compendium of Chemical Terminology*, 5th ed. International Union of Pure and Applied Chemistry; 2025. Online version 5.0.0, 2025. <https://doi.org/10.1351/goldbook.P04859>



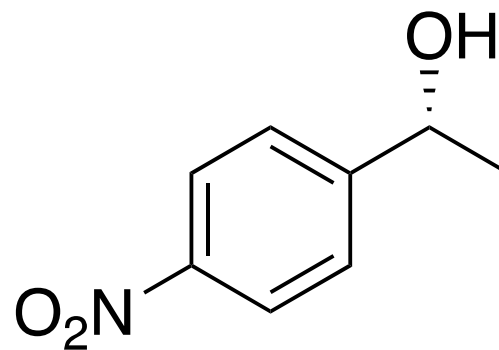
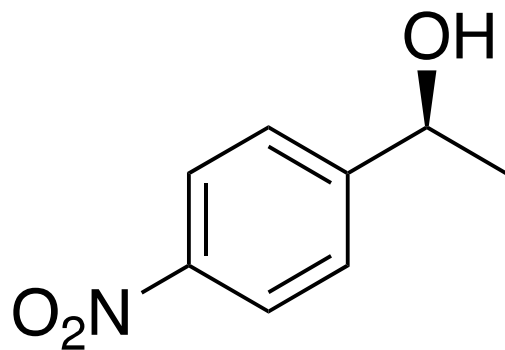
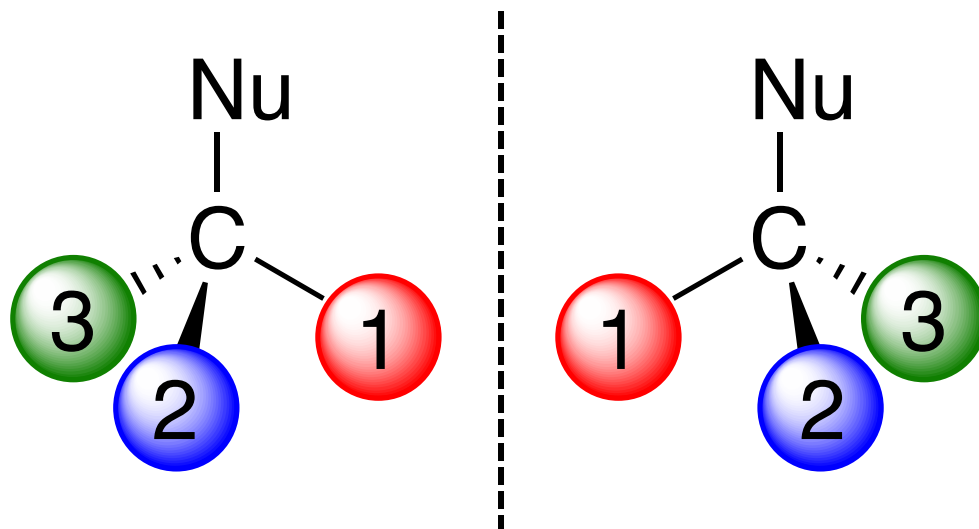
Para nombrar las caras se siguen las mismas reglas de prioridad de Cahn-Ingold-Prelog que se usan para designar enantiómeros *R* o *S*.



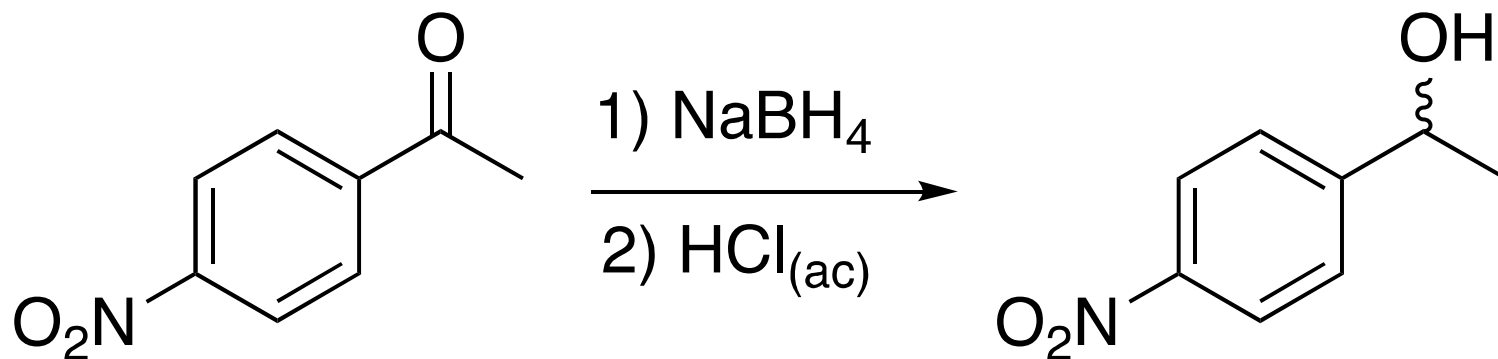




# Generación de enantiómeros



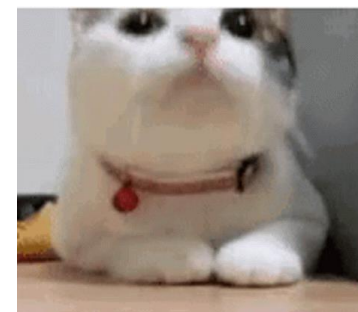
Considere la siguiente reacción:



¿Es quimioselectiva?



¿Es regioselectiva?

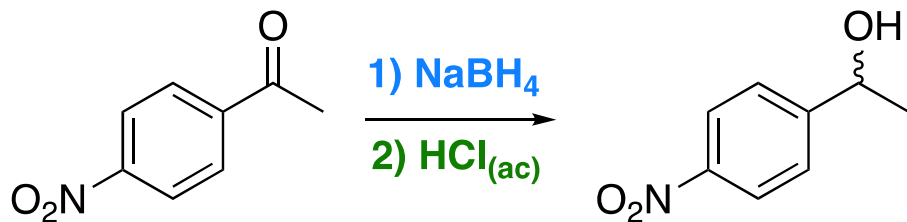
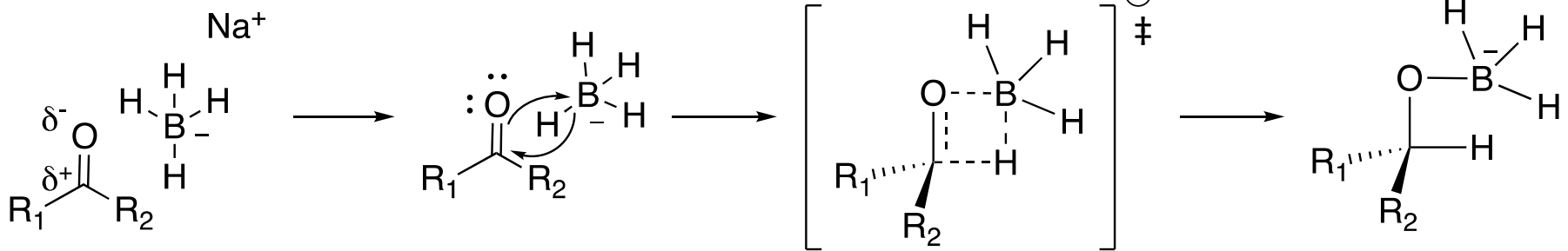


¿Es estereoselectiva?



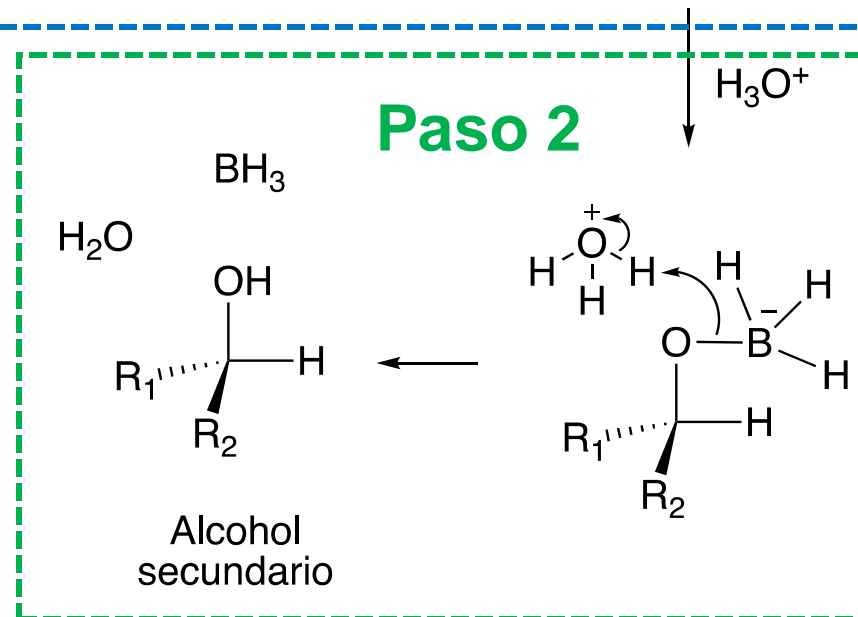
# Mecanismo

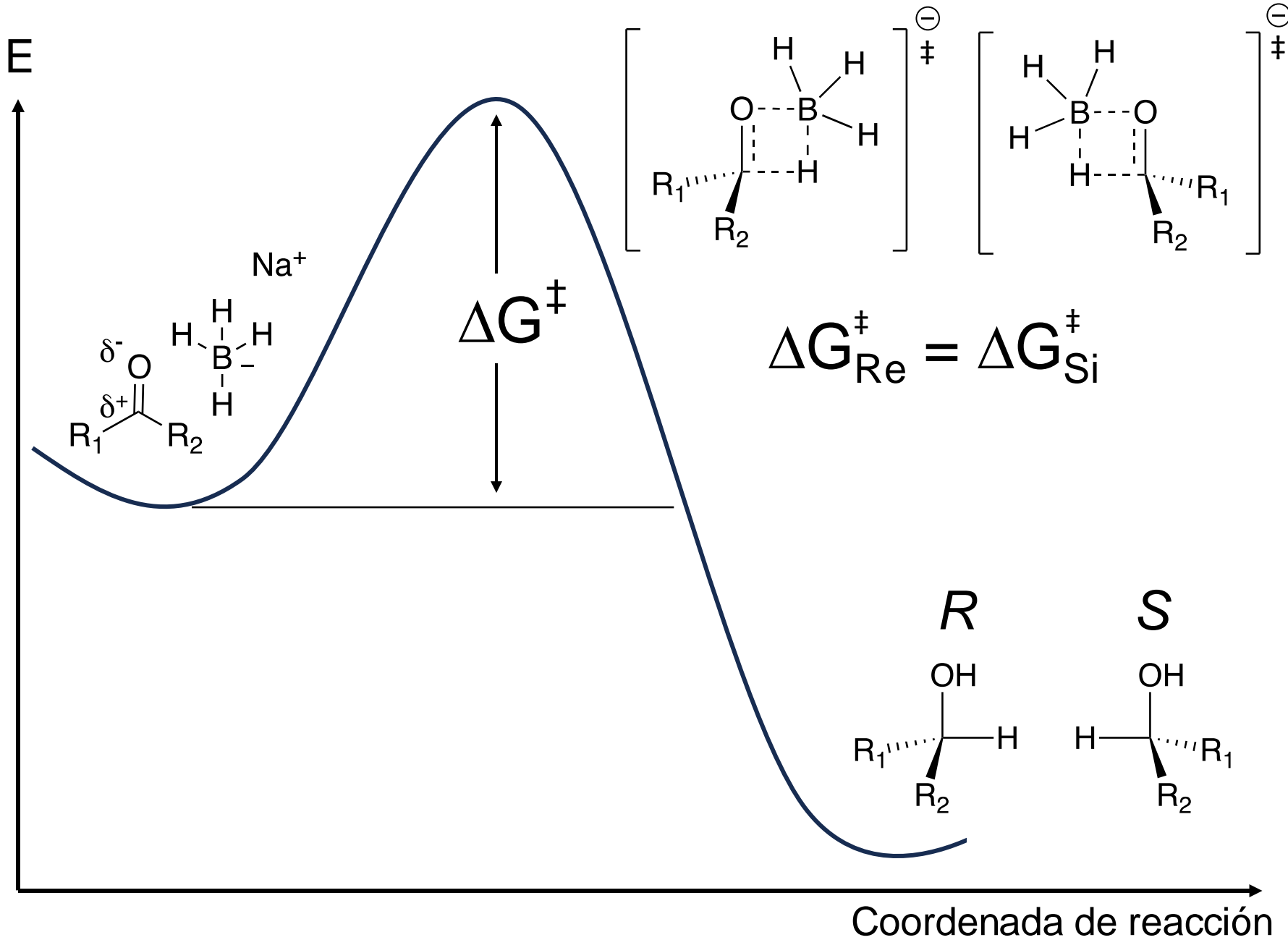
## Paso 1



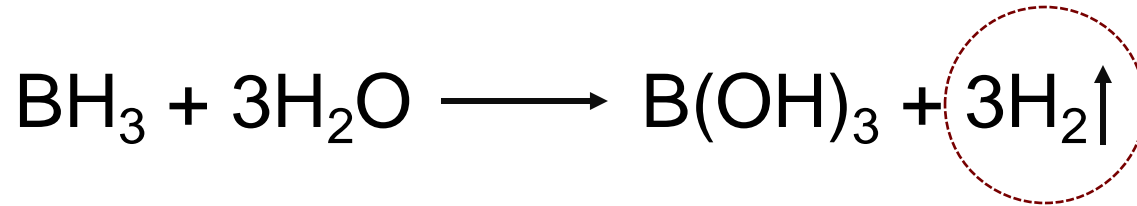
Carey, F. A.; Sundberg, R. J. *Advanced Organic Chemistry: Part A: Structure and Mechanisms*; Springer Science & Business Media, **2007**.

## Paso 2

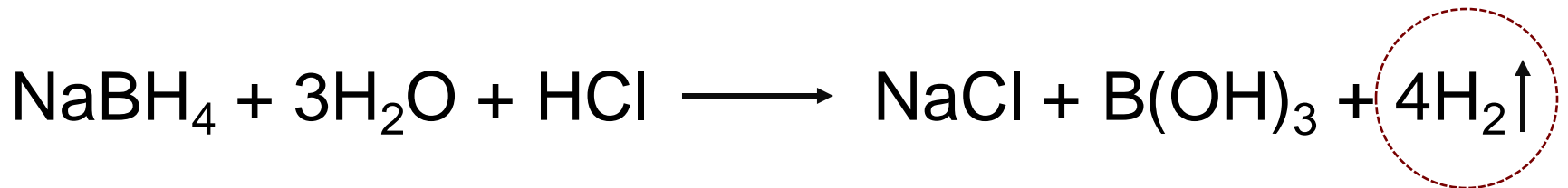




## Hidrólisis del borano generado



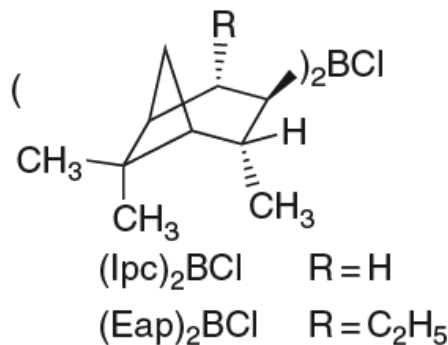
## Hidrólisis del borohidruro de sodio remanente



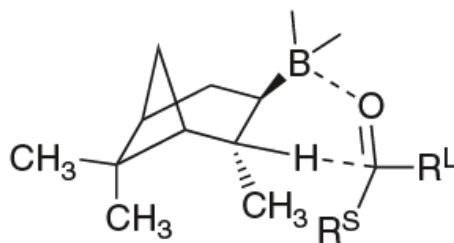
Netskina, O. V; Filippov, T. N.; Komova, O. V; Simagina, V. I. Hydrogen Generation by Both Acidic and Catalytic Hydrolysis of Sodium Borohydride. **2018**, 5 (1), 41–48.

¿Es posible realizar reducciones enantioselectivas?

Sí, con un agente reductor quirral.



In most cases, the enantioselectivity can be predicted by a model that places the smaller carbonyl substituent toward the isopinocampheyl methyl group.<sup>149</sup>

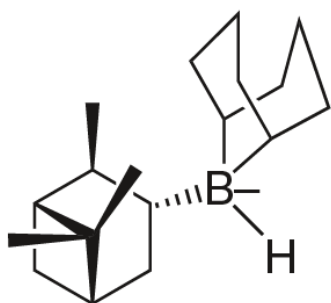


Carey, F. A.; Sundberg, R. J. *Advanced Organic Chemistry: Part A: Structure and Mechanisms*; Springer Science & Business Media, **2007**.

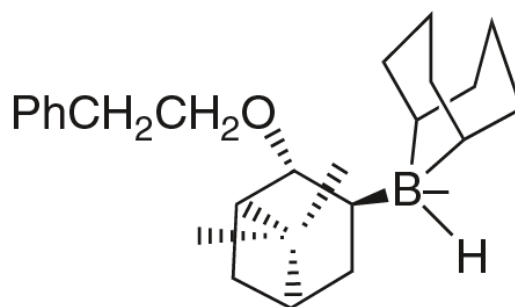
**Table 2.6. Enantioselective Reduction of Ketones**

Reagent	Ketone	% e.e.	Configuration
Alpine-Borane <sup>©a</sup>	3-Methyl-2-butanone	62	<i>S</i>
NB-Enantride <sup>©b</sup>	2-Octanone	79	<i>S</i>
Eapine-Hydride <sup>c</sup>	2-Octanone	78	<i>S</i>
(Ipc) <sub>2</sub> BCl <sup>d</sup>	2-Acetylnaphthalene	94	<i>S</i>
(Ipc)( <i>t</i> -Bu)BCl <sup>e</sup>	Acetophenone	96	<i>R</i>
(Ipc) <sub>2</sub> BCl <sup>f</sup>	2,2-Dimethylcyclohexanone	91	<i>S</i>
(Eap) <sub>2</sub> BCl <sup>g</sup>	3-Methyl-2-butanone	95	<i>R</i>

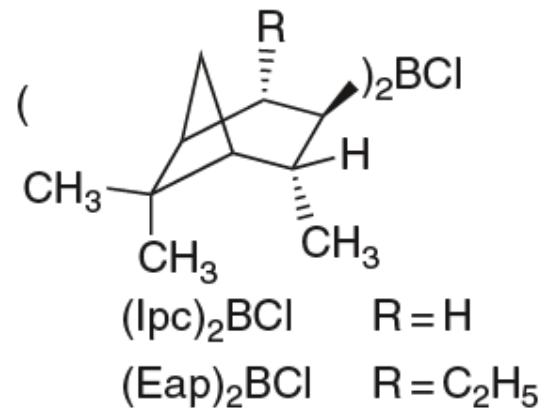
NOTE: © Trademark of Sigma Aldrich Corporation.



Alpine-Hydride\*



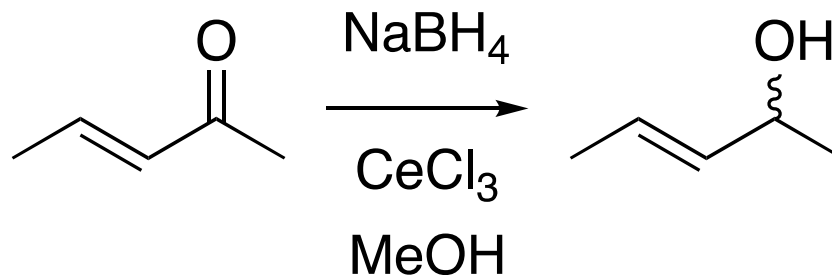
NB-Enantride\*



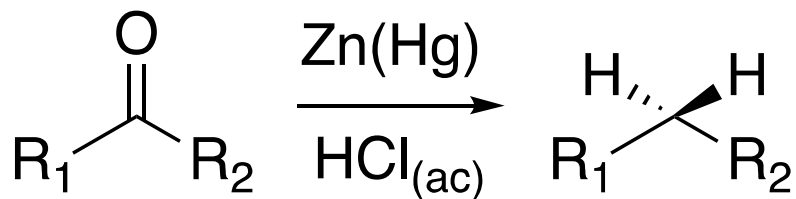
Carey, F. A.; Sundberg, R. J. *Advanced Organic Chemistry: Part A: Structure and Mechanisms*; Springer Science & Business Media, **2007**.

# Otras reducciones

## Reducción de Luche



## Reducción de Clemmensen



## Reducción de Wolff-Kishner

