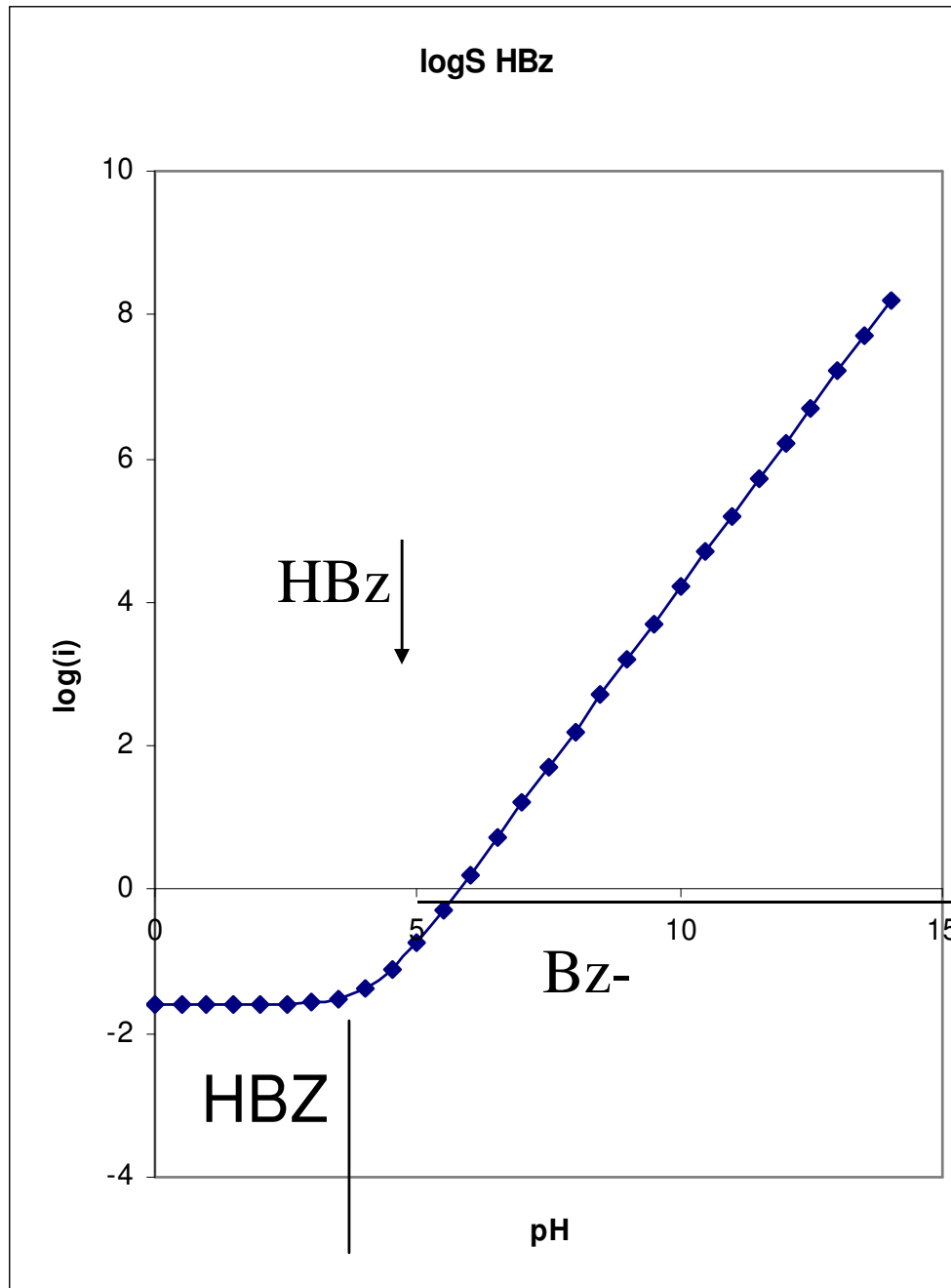


$$\log S' = \log S_0 + \log \left[1 + 10^{-pK_a + pH} \right]$$

$$\log S' = \log(0.0264) + \log \left[1 + 10^{-4.19 + pH} \right]$$

$$\log S' = -1.6 + \log \left[1 + 10^{-4.19 + pH} \right]$$



FQ UNAM Alejandro Baeza