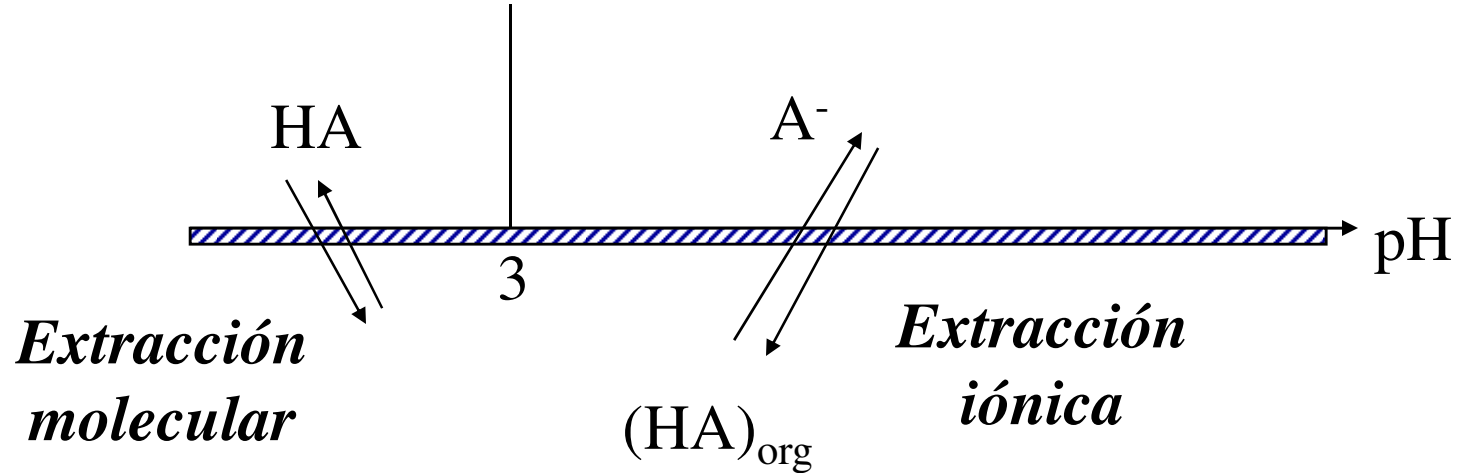


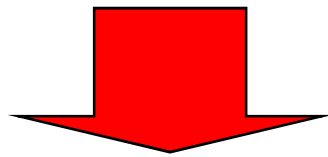
# Acido salisilico DUZP (combinado)



$$\text{Log} (f/1-f) = f(\text{pH})$$

$$\frac{\text{HA}}{\text{In CoVo}} = \frac{(HA)_{org}}{\text{Eq CoVo}(1-f) + f\text{CoVo}}$$

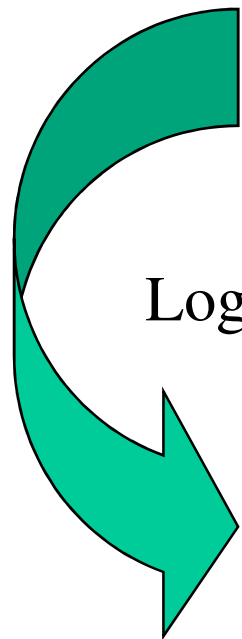
$$K_D = 76.92; \log K = 1.9$$



*continua*

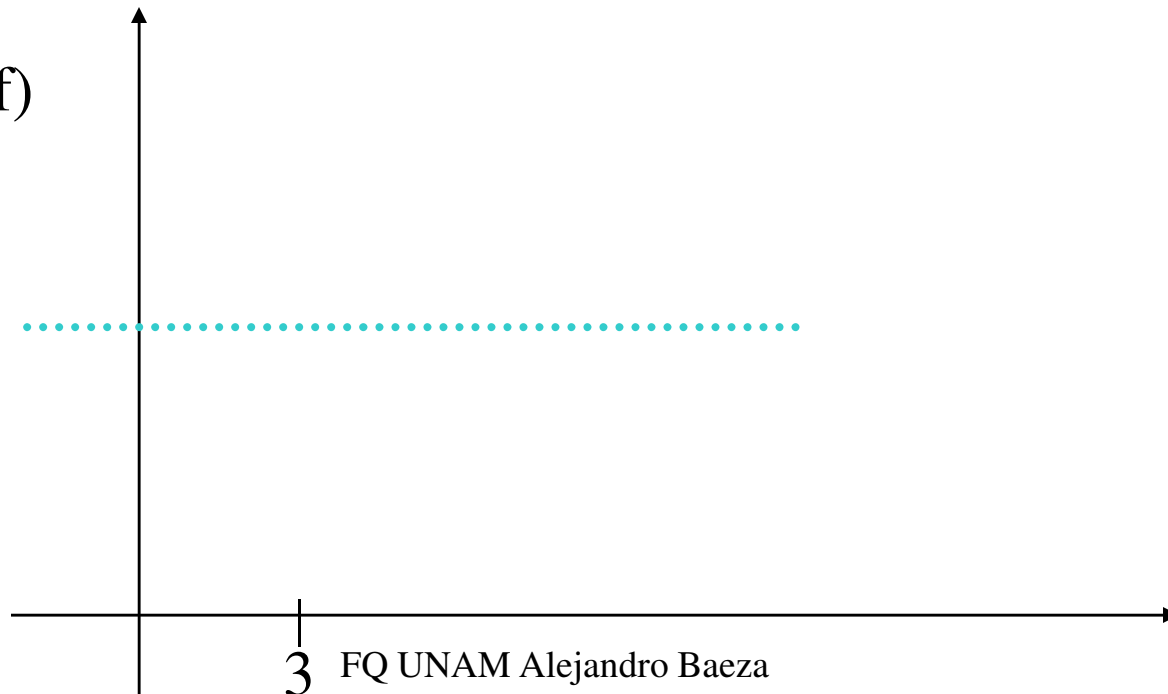
$$K_D = \frac{(HA)_{org}}{(HA)} = \frac{fC_0V_o/V_{org}}{C_0V_o(1-f)/V_o} = \frac{f(Vac/V_{org})}{(1-f)}$$

$$\text{Log}(f/1-f) = \text{log } K_D + \text{log}(V_{org}/Vac) = 2 + 1 = 3$$



Log(f/1-f)

3

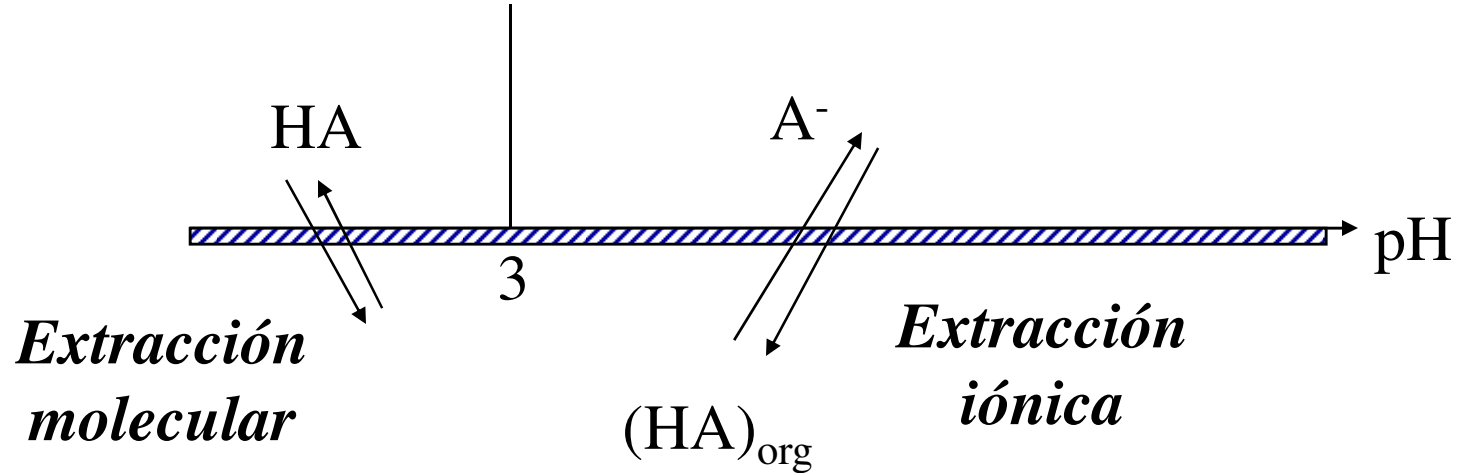


3

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pH 2

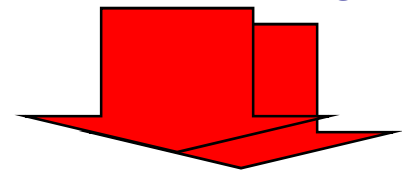
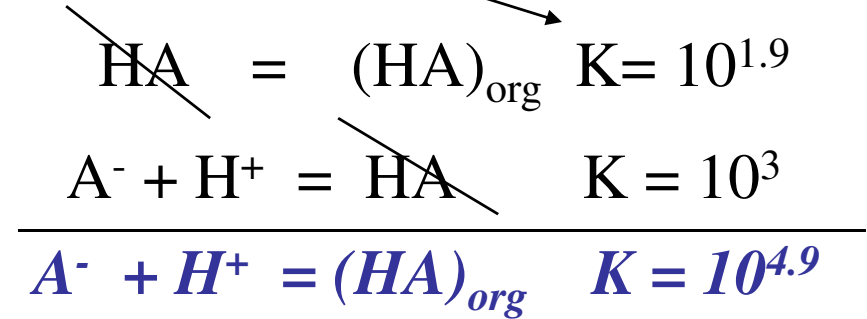
# Acido salisilico DUZP (combinado)



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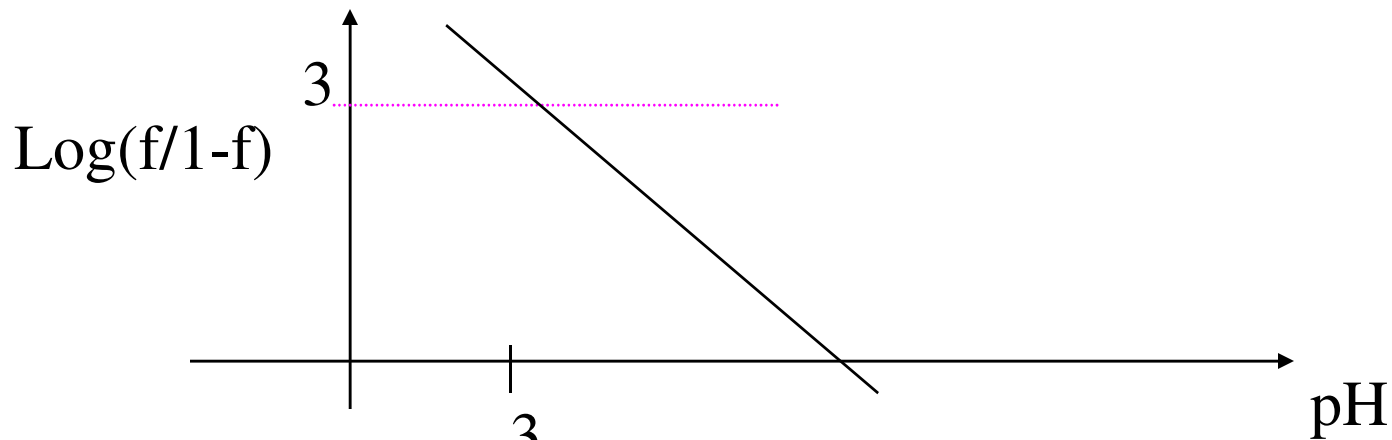


In Eq

$$\frac{fCoVac}{CoVac(1-f)} \cdot 10^{-pH} = \frac{fCoVac}{fCoVac}$$

$$K_E = \frac{\cancel{fCoVac}/Vorg}{\cancel{CoVac}(1-f)/Vac \cdot (10^{-pH})} = \frac{f(Vac/Vorg)}{(1-f) 10^{-pH}}$$

$$\text{Log}(f/1-f) = \text{log } K + \text{log}(Vorg/Vac) + pH = 5.9 - pH$$



$$\text{pH} = \text{pKa}$$

$$(\text{HA})_{\text{ac}} = C_0/2$$

$$(\text{HA})_{\text{org}} = C_0/2$$

$$\text{Log} (f/1-f) = \text{log} C_0 - 0.3$$

$$\text{Log} (f/1-f) = (f/1-f)$$

$$(f/1-f) = 1$$

$$f = 0.5$$

