

CURSO DE TRANSFERENCIA DE MASA

DGAPA / PAPIME / PEI 103325

(1649)

CONCEPTOS DE FLUJO, FLUX Y VELOCIDAD

REFERENCIAS CURSO TRANSFERENCIA DE MASA

- 1) Introductory Transport Phenomena.**
R. Byron Bird, Warren E. Stewart, Edwin N. Lightfoot, Daniel J. Klingenberg.
JohnWiley & Sons, Inc., 2015.
- 2) Diffusion. Mass Transfer In Fluid Systems.**
E. L. Cussler
Cambridge University Press, UK, 2009, 3rd ed.
- 3) Mass Transfer In Multicomponent Mixtures.**
J. A. Wesselingh, R. Krishna.
Delft University Press, Netherlands, 2000.
- 4) Multicomponent Mass Transfer.**
R. Taylor, R. Krishna.
John Wiley & Sons, Inc., USA, 1993.
- 5) Principles Of Mass Transfer And Separation Processes.**
B. K. Dutta.
PHI Learning Private Limited, New Delhi India, 2007.
- 6) Diffusion And Mass Transfer.**
J. S. Vrentas, C. M. Vrentas.
Taylor and Francis, USA, 2013.
- 7) Estimation of Diffusivities**
Dr. K. Wegner.
ETH Zürich, Lecture 8, 08.11.2017.
- 8) Fenómenos De Transporte. Aplicaciones Con Métodos Numéricos.**
Luis Carrasco Venegas.
Empresa Editora Macro, 2011.
- 9) Mass Transfer. Principles And Applications.**
D. Basmadjian.
CRC Press, 2013.
- 10) Absorption And Stripping.**
P. Chattopadhyay.
Asian Books Private Limited, New Delhi, India, 2007.
- 11) Packed Towers In Processing And Environmental Technology.**
R. Billet.
VCH Verlagsgesellschaft mbH, Weinheim, Federal Republic of Germany, 1995.
- 12) The Properties of Gases and Liquids.**
B. E. Poling, J. M. Prausnitz, John P. O'Connell.
The McGraw-Hill Companies, Inc., 2001.
- 13) Introducción A La Mecánica Del Medio Continuo.**
A. Ortiz Prado, J. A. Ortiz Valera, Osvaldo Ruiz Cervantes.
UNAM, Facultad de Ingeniería, 2013.
- 14) Separation Process Engineering. Includes Mass Transfer Analysis.**
P. C. Wankat.
Pearson Education, Inc., USA, 2012, third edition.
- 15) Mass Transfer Operations for the Practicing Engineer.**

Louis Theodore, Francesco Ricci.

John Wiley and Sons, USA, 2010.

16) Packed Tower Design and Applications. Random and Structured Packings.

Ralph F. Strigle Jr.

Gulf Publishing Co., 1994, 2nd edition.

17) Mass Transfer In Chemical Engineering Processes.

Jozef Markos

InTech, 2011.

18) Mass Transfer. From Fundamentals To Modern Industrial Applications.

Koichi Asano.

Wiley-VCH, 2006.

19) Mass Transfer in Multiphase Systems And Its Applications.

Mohamed El-Amin.

InTech, 2011

20) Mass Transfer Operations.

Robert E. Treybal.

McGraw-Hill, 3rd Edition, 1980.

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INCORPORACION DE TIC'S PARA LA ENSEÑANZA Y EVALUACION DEL DESEMPEÑO DE LOS ALUMNOS EN ASIGNATURAS COMPARTIDAS DE DOS CARRERAS DE LA FACULTAD DE QUIMICA.



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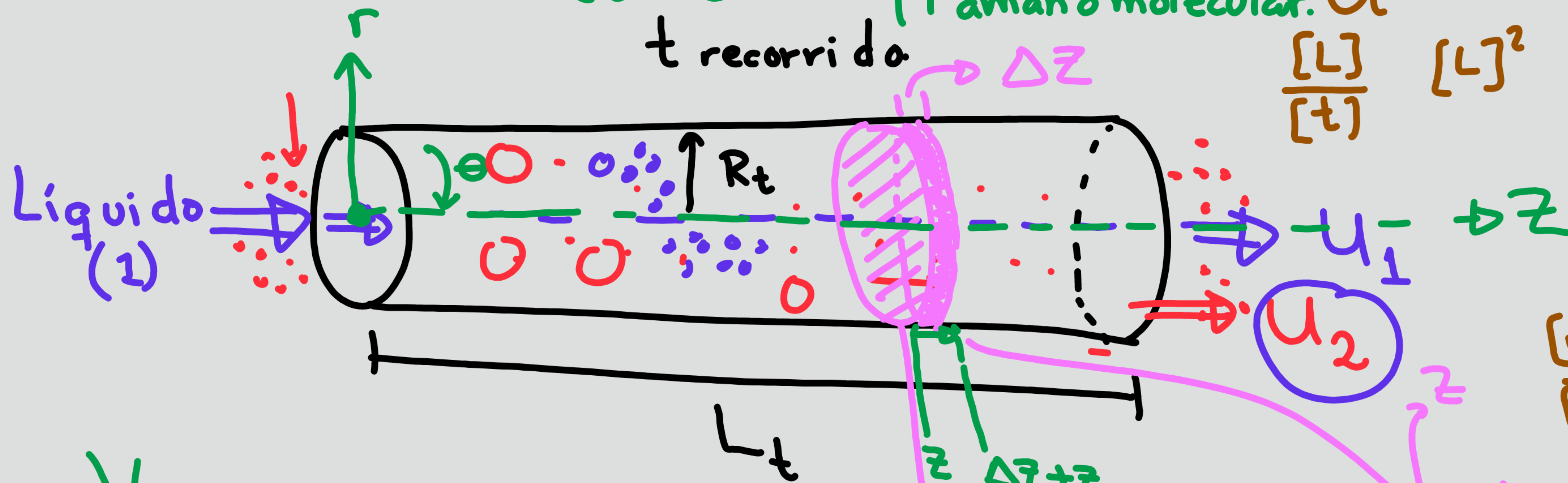
Flujo, Flux, Velocidad.

Convectivo { velocidad fluidos.
t recorrido tamaño molecular.

$$U \cdot A = Q$$

$$\frac{[L]}{[t]} [L]^2 = \frac{[L]^3}{[t]} [A]$$

$$U_1 \neq U_2$$



$$F_m \cdot \frac{1}{M_n} = F$$

$$\frac{[m]}{[t]} \frac{[mol]}{[m]} M_n = \frac{[mol]}{[t]}$$

$$V_{control} = A_s \cdot \Delta z$$

$$V_{control} = \pi (R_t)^2 \cdot \Delta z$$

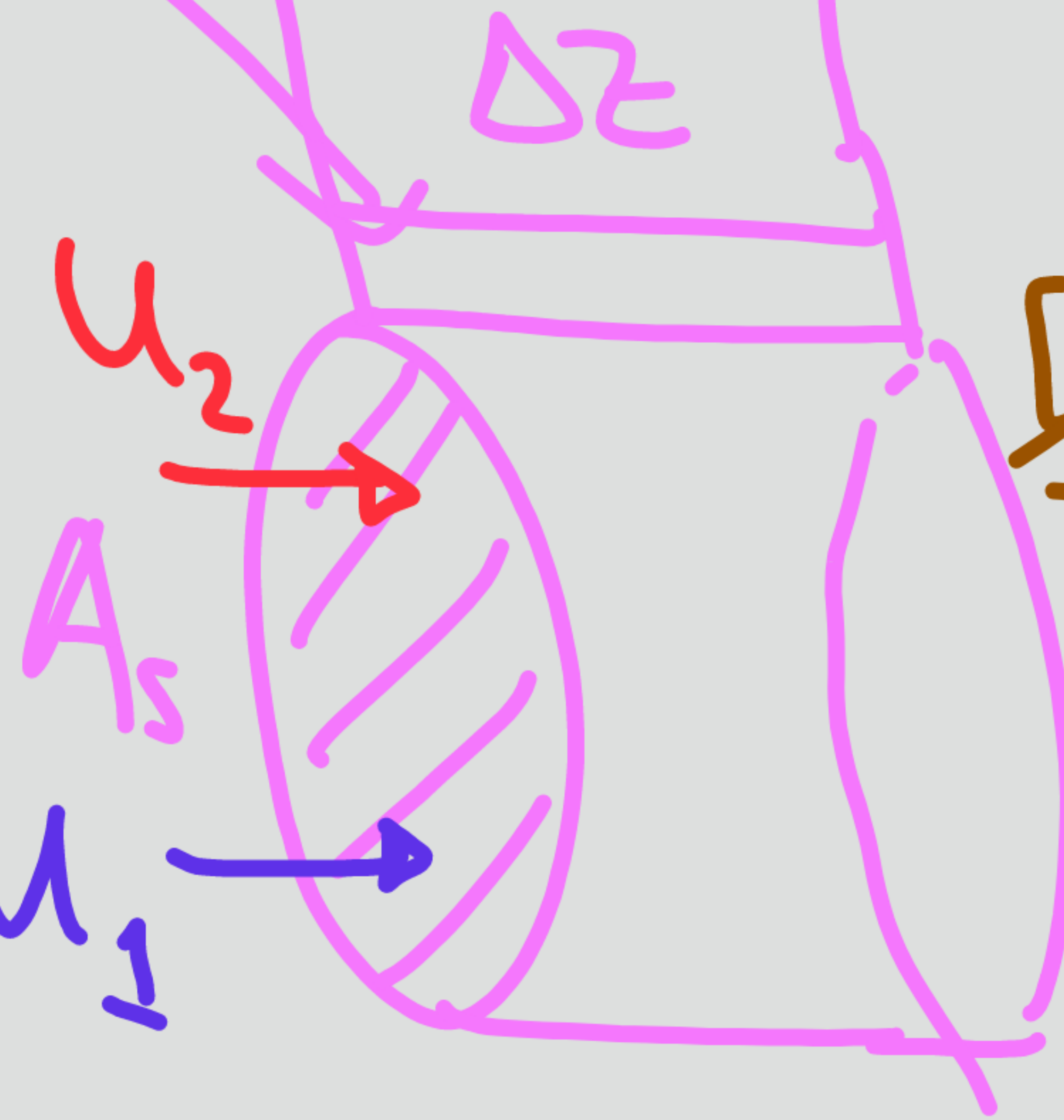
$$A_s = \pi (R_t)^2$$

$$\frac{K}{t} \cdot \frac{mol}{L^2} \cdot \frac{mol}{mol}$$

$$\frac{[mol]}{[t][L]^2} = J = \frac{F}{A} = U \cdot \rho \cdot \frac{t}{M_n} U_1$$

$$Q \cdot \rho = F_m$$

$$\frac{[L]^3}{[t]} \frac{[m]}{[L]^3} = \frac{[m]}{[t]}$$



[másico]

$\frac{[masa]}{[L]^3[t]}$

→ flux

n_i

másico

ρ_i → densidad

del componente "i"

$$u_i = \frac{n_i}{\rho_i}$$

↓
velocidad
en términos
másicos del
componente "i"

fracción masa de "i"

$$V = \sum_{i=1}^{N.C.} (w_i u_i)$$

↓
velocidad
total del fluido
en términos másicos

[molar]

$$u_i = \frac{n_i}{\rho_i} \cdot \frac{\bar{M}_i}{M_i} = \frac{N_i}{C_i}$$

N_i → Flux molar

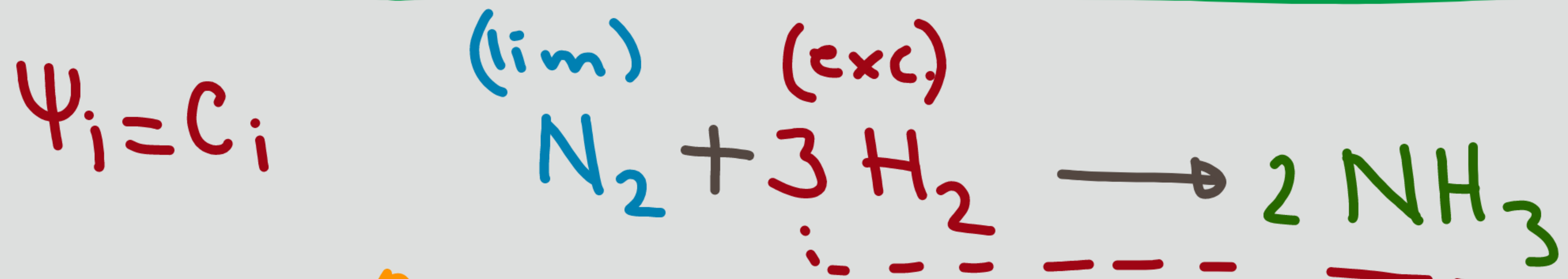
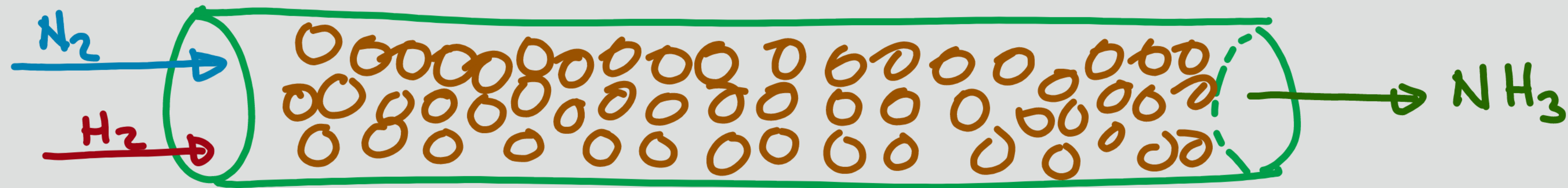
C_i → Conc. del comp. "i"

velocidad total del fluido molar

$$U = \sum_{i=1}^{N.C.} (x_i u_i)$$

$$r = - \boxed{k_c'} C_{N_2} \boxed{C_{H_2}^3} = - k_c'' C_{N_2}$$

↪ pseudo constante



$$\frac{\partial C_{H_2}}{\partial t} + \nabla \cdot (\underline{v} \cdot C_{H_2}) = \nabla \cdot (D \nabla C_{H_2}) + 3 (-k_c'' C_{N_2})$$

$$\frac{\partial C_{N_2}}{\partial t} + \nabla \cdot (\underline{v} \cdot C_{N_2}) = \nabla \cdot (D \nabla C_{N_2}) + (-k_c'' C_{N_2})$$

[másico]

$$n_M = \sum_{i=1}^{N.C.} n_i$$

$$n_M = \sum_{i=1}^{N.C.} \frac{F_{m,i}}{A}$$

flujo másico de "i"

$$n_M = \frac{1}{A} \sum_{i=1}^{N.C.} F_{m,i}$$

Flux másico total.

[molar]

$$N_M = \sum_{i=1}^{N.C.} N_i$$

$$N_M = \sum_{i=1}^{N.C.} \frac{F_i}{A}$$

$$N_M = \frac{1}{A} \sum_{i=1}^{N.C.} F_i$$

Flux molar de "i"

Flux molar total.

[máscico]

$$\begin{aligned} n_M &= \sum_{i=1}^{N.C.} (\rho_i u_i) \\ &= \sum_{i=1}^{N.C.} (\rho_M w_i \cdot u_i) \quad \text{densidad mezcla.} \\ &= \rho_M \sum_{i=1}^{N.C.} (w_i u_i) \\ &= \rho_M V \end{aligned}$$

[molar]

$$\begin{aligned} N_M &= \sum_{i=1}^{N.C.} (C_i u_i) \\ &= \sum_{i=1}^{N.C.} (C_M x_i \cdot u_i) \quad \text{Conc. total de la mezcla.} \\ &= C_M \sum_{i=1}^{N.C.} (x_i \cdot u_i) \\ &= C_M U \end{aligned}$$

T=cte Difusión

$$C_1 = \frac{\bar{n}_1}{V_2} > 0$$

$$C_1 = \frac{\bar{n}_1^{t=0}}{V_2} = 0$$

$t < 0$ gases separados
 $t = 0$ abro la válvula.

válvula

$$\bar{n}_1 = \text{moles (1)}$$

$$M_1 \sim M_2 \quad (2)$$

$$\bar{n}_2 = \text{moles}$$

$$C_2 = \frac{\bar{n}_2^{t=0}}{V_1} = 0$$

$$V_1 = V_2$$

$$T$$

$$C_2 = \frac{\bar{n}_2}{V_2} > 0$$

$$\nabla C_1$$

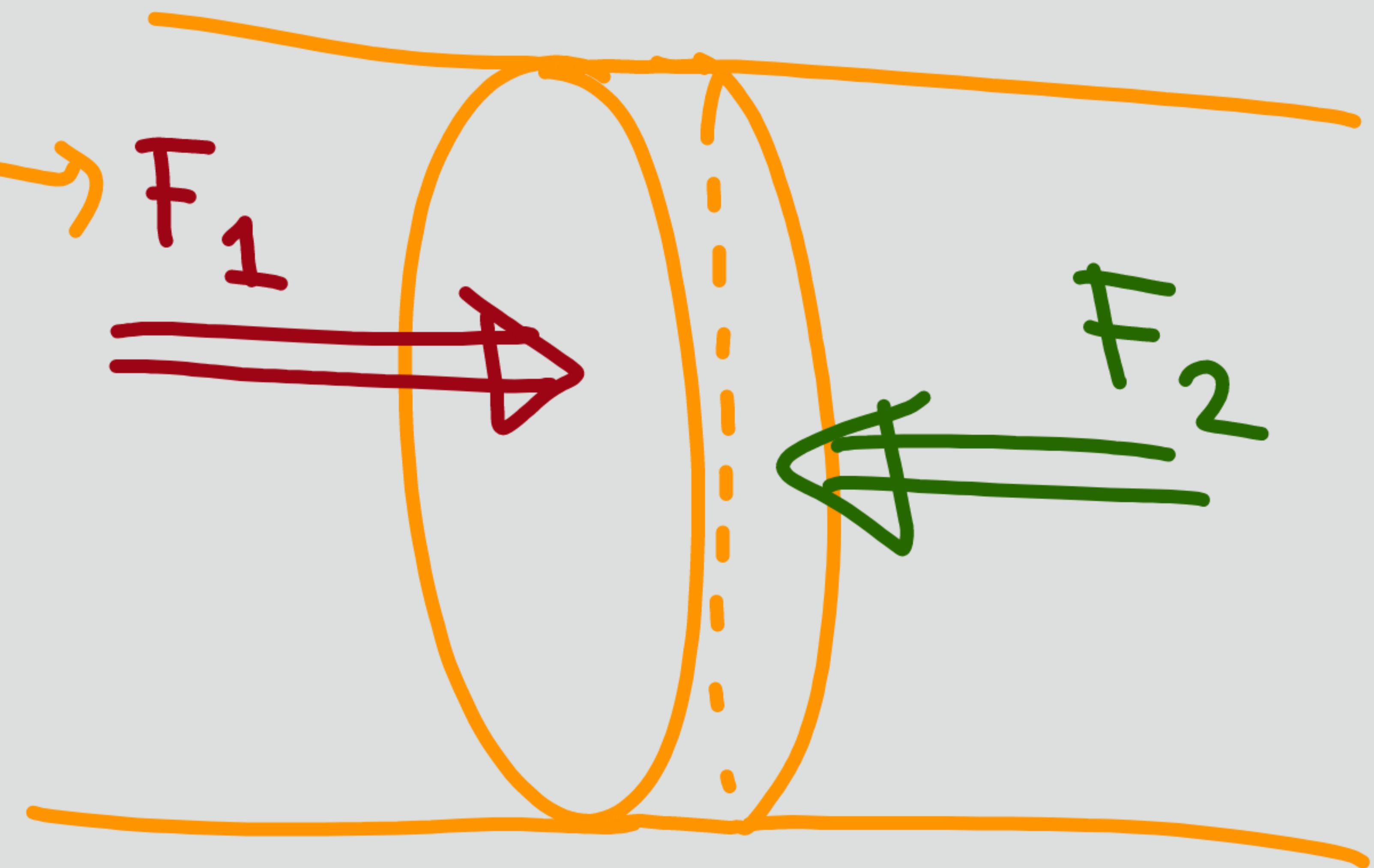
$$\nabla C_2$$

Gas 1

Gas 2

$$P_1 \sim$$

$$P_2$$



$$J_1 = \frac{F_1}{A}$$

↓
Flux
difusivo
molar
de "1"

$$J_1 \propto \nabla C_1$$

$$J_1 = -D_{1/2} \nabla C_1$$

$$J_2 = \frac{F_2}{A}$$

↓
Flux
molar
difusivo
de "2"

$$J_2 \propto \nabla C_2$$

$$J_2 = -D_{2/1} \nabla C_2$$

D_{ij} = coeficiente
de difusión
binario de "i"
tratando de
penetrar un
campo de
moléculas "j"