

Table 28: List of Fe species

FACT	SUPCRT	LLNL	JNC-TDB
$\text{Fe}[2+]$ , $\text{Fe}[3+]$ , $\text{FeOH}[+]$ , $\text{FeOH}[2+]$ , $\text{HFeO}_2[-]$ , $\text{Fe}(\text{OH})_2[+]$ , $\text{FeO}_2[2-]$ , $\text{Fe}(\text{OH})_3(\text{aq})$ , $\text{Fe}_2(\text{OH})_2[4+]$ , <b>Fe(s)</b> , <b>Fe(s2)</b> , <b>Fe<sub>2</sub>O<sub>3</sub>(s2)</b> , <b>Fe<sub>2</sub>O<sub>3</sub>(s3)</b> , <b>Fe<sub>3</sub>O<sub>4</sub>(s2)</b> , <b>Fe<sub>3</sub>O<sub>4</sub>(s3)</b> , <b>Fe<sub>3</sub>O<sub>4</sub>(s4)</b> , $\text{Fe}(\text{OH})_2(\text{s})$ , $\text{Fe}(\text{OH})_3(\text{s})$ , $\text{Fe}_2\text{O}_3(\text{H}_2\text{O})(\text{s})$ , <b>FeO(s)</b> , <b>Fe<sub>2</sub>O<sub>3</sub>(s)</b> , <b>Fe<sub>3</sub>O<sub>4</sub>(s)</b>	$\text{HFeO}_2(\text{aq})$ , $\text{FeO}_2[-]$ , $\text{FeO}(\text{aq})$ , $\text{Fe}[2+]$ , $\text{Fe}[3+]$ , $\text{FeOH}[+]$ , $\text{FeOH}[2+]$ , $\text{HFeO}_2[-]$ , $\text{FeO}[+]$ , <b>FeO</b> <b>(Ferrous-oxide)</b> , <b>Fe<sub>2</sub>O<sub>3</sub>(Hematite)</b> , <b>Fe<sub>3</sub>O<sub>4</sub>(Magnetite)</b>	$\text{Fe}[2+]$ , $\text{Fe}[3+]$ , $\text{Fe}(\text{OH})_2(\text{aq})$ , $\text{Fe}(\text{OH})_2[+]$ , $\text{Fe}(\text{OH})_3(\text{aq})$ , $\text{Fe}(\text{OH})_3[-]$ , $\text{Fe}(\text{OH})_4[-]$ , $\text{Fe}_2(\text{OH})_2[4+]$ , $\text{Fe}_3(\text{OH})_4[5+]$ , $\text{FeOH}[+]$ , $\text{FeOH}[2+]$ , <b>Fe(OH)<sub>2</sub>(ppd)</b> , <b>Fe(OH)<sub>3</sub>(ppd)</b> , <b>FeO(c)</b> <b>Goethite</b> , <b>Hematite</b> , <b>Magnetite</b> , <b>Wustite</b>	$\text{Fe}[2+]$ , $\text{Fe}[3+]$ , $\text{FeOH}[2+]$ , $\text{Fe}(\text{OH})_2[+]$ , $\text{Fe}(\text{OH})_3(\text{aq})$ , $\text{FeOH}_4[-]$ , $\text{Fe}_2(\text{OH})_2[4+]$ , $\text{Fe}_3(\text{OH})_4[5+]$ , $\text{FeOH}[+]$ , $\text{Fe}(\text{OH})_2(\text{aq})$ , $\text{Fe}(\text{OH})_3[-]$ , <b>Fe(OH)<sub>3</sub>(a)</b> <b>Hematite</b> , <b>Magnetite</b> , <b>Goethite</b>

HATCHES			
$\text{Fe}[2+]$ , $\text{Fe}[3+]$ , $\text{Fe}(\text{OH})[+]$ , $\text{Fe}(\text{OH})_2(\text{aq})$ , $\text{Fe}(\text{OH})_2[+]$ , $\text{Fe}(\text{OH})_3(\text{aq})$ , $\text{Fe}(\text{OH})_3[-]$ , $\text{Fe}(\text{OH})_4[-]$ , $\text{Fe}(\text{OH})_4[2-]$ , $\text{Fe}_2(\text{OH})_2[4+]$ , $\text{Fe}_3(\text{OH})_4[5+]$ , $\text{FeOH}[2+]$ , <b>Fe(OH)<sub>2</sub></b> , <b>Fe(OH)<sub>3</sub></b> , <b>FeO(c)</b> , <b>Goethite</b> , <b>Hematite</b> , <b>Magnetite</b> , <b>Wustite</b>			

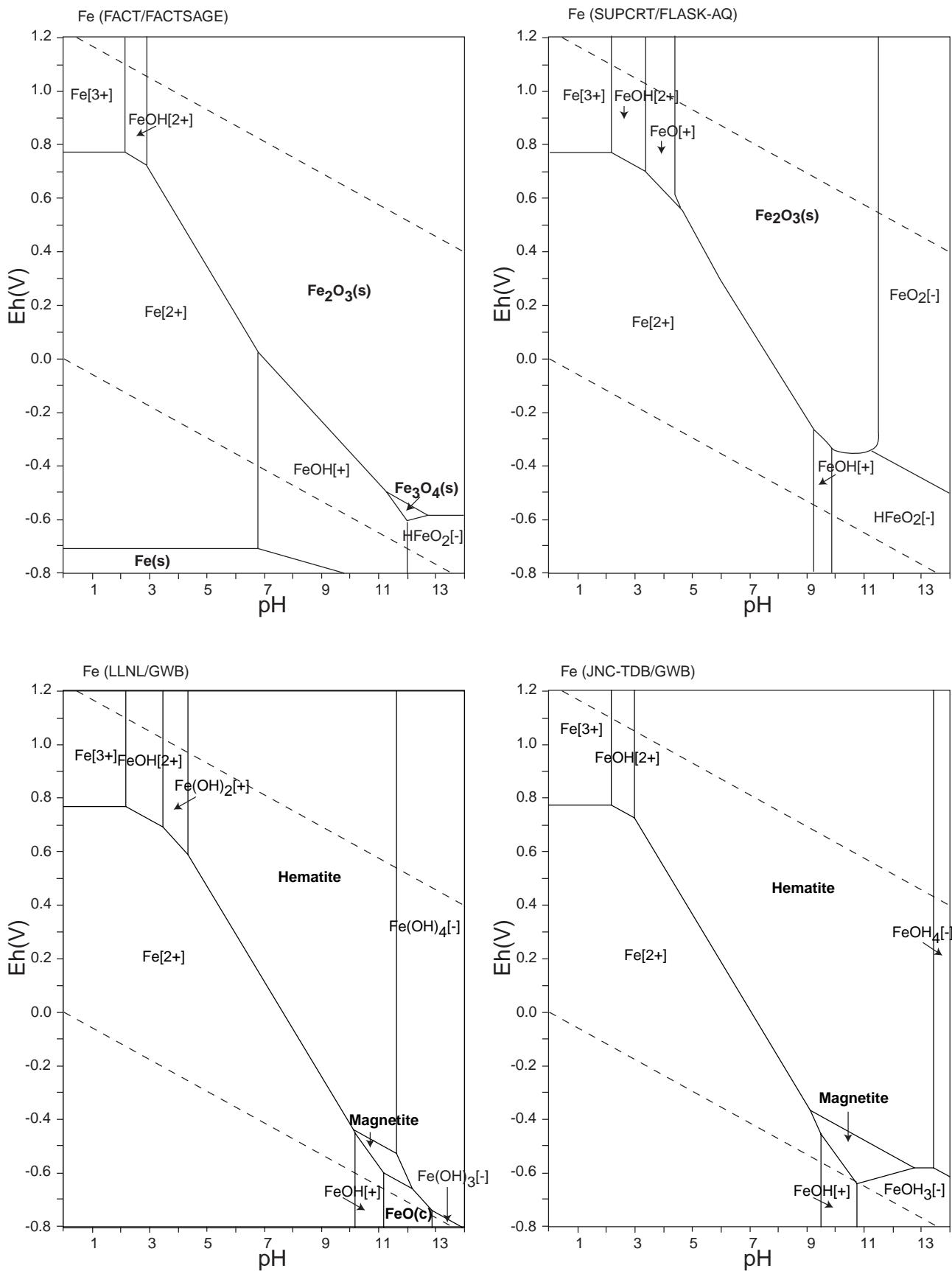


Figure 47: Eh-pH diagrams of the system Fe-O-H (1).  $\sum \text{Fe} = 10^{-10}$ , 298.15K,  $10^5$  Pa.

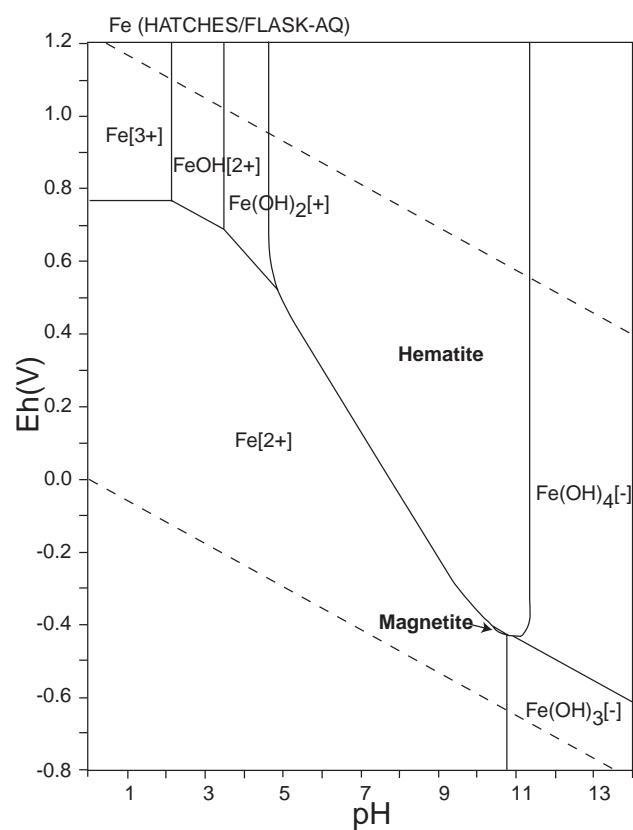


Figure 48: Eh-pH diagrams of the system Fe-O-H (2).  $\sum \text{Fe} = 10^{-10}$ , 298.15K,  $10^5$  Pa.