

Estados de oxidación más representativos

| Grupo 1 (I A) | | Grupo 2 (II A) | | Grupo 13 (III A) | | Grupo 14 (IV A) | | Grupo 15 (V A) | | Grupo 16 (VI A) | | Grupo 17 (VII A) | |
|---------------|--------|----------------|----|------------------|--------|-----------------|------------|----------------|----------------|-----------------|------------|------------------|--------------------|
| * H | 1+, 1- | | | | | | | | | | | | |
| Li | 1+ | Be | 2+ | * B | 3+ | * C | 2+, 4+, 4- | * N | 1+, 3+, 5+, 3- | * O | 1-, 2- | * F | 1- |
| Na | 1+ | Mg | 2+ | Al | 1+, 3+ | * Si | 2+, 4+, 4- | * P | 1+, 3+, 5+, 3- | * S | 4+, 6+, 2- | * Cl | 1+, 3+, 5+, 7+, 1- |
| K | 1+ | Ca | 2+ | Ga | 1+, 3+ | Ge | 2+, 4+ | * As | 1+, 3+, 5+, 3- | * Se | 4+, 6+, 2- | * Br | 1+, 3+, 5+, 7+, 1- |
| Rb | 1+ | Sr | 2+ | In | 1+, 3+ | Sn | 2+, 4+ | Sb | 3+, 5+ | * Te | 4+, 6+, 2- | * I | 1+, 3+, 5+, 7+, 1- |
| Cs | 1+ | Ba | 2+ | Tl | 1+, 3+ | Pb | 2+, 4+ | Bi | 3+, 5+ | Po | 4+, 6+ | At | |
| Ff | | Ra | | | | | | | | | | | |

| Grupo 4 (IV B) | |
|-----------------|----------------|
| Ti | 4+ |
| Grupo 5 (V B) | |
| V | 5+ |
| Grupo 6 (VI B) | |
| Cr | 2+, 3+, 6+ |
| Grupo 7 (VII B) | |
| Mn | 2+, 4+, 6+, 7+ |

| Grupo 8, 9 y 10 (VIII B) | | | | | |
|--------------------------|--------|----|--------|----|----|
| Fe | 2+, 3+ | Pd | 2+, 4+ | Os | 4+ |
| Co | 2+, 3+ | Pt | 2+, 4+ | Ir | 4+ |
| Ni | 2+ | | | | |

| Grupo 11 (1 B) | Grupo 11 (1 B) |
|----------------|----------------|
| Cu | 1+, 2+ |
| Ag | 1+ |
| Au | 1+, 3+ |
| Zn | 2+ |
| Cd | 2+ |
| Hg | 1+, 2+ |

* No Metáles

Radioactive